AQA Chemistry triple Paper 1 2023

Why is it called the periodic table? - ANS-because the word "periodic" means similar properties occurring at regular intervals and they are grouped by their similar properties

How many columns are in the periodic table? - ANS-18 columns that are grouped into 1,2,3,4,5,6,7,0 and transition metals in the middle

How are elements arranged on the periodic table? - ANS-In order of increasing atomic number. Left to the "stairs" contain metals, right to the "stairs" contain nonmetals.

What is Döbreiners Triads? - ANS-- scientist who noticed elements had similar chemical properties, they occurred in threes.

 these groups were called "triads", an example would be Lithium, Sodium and Potassium which all react rapidly with water

What are Newlands' Octaves? - ANS-- 40 years after Döbreiner, Newland noticed that after he arranged elements in increasing atomic weight, every 8th element had similar properties (eg. Lithium, sodium and potassium)

-called law of octaves

-problems with this, by always sticking to exact order of weight some elements were grouped together that had totally different properties

Who developed the first periodic table? - ANS--Dmitri Mendeleev in 1869

-started arranging elements in order of atomic mass

- would switch order of specific elements so they fit patterns of others in the group

left gaps for undiscovered elements

-predicted properties of undiscovered ones so scientists accepted his one as correct

Differences between the first periodic table and the modern periodic table - ANSarranged in order of atomic number (number of protons) this is because they weren't
discovered at his time so he ordered them in order of atomic weight

- atomic weight method is faulty as elements can appear in wrong order due to isotopes

modern periodic table has group 0 , noble gases

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