

5 (f) Table 3 is repeated to help you answer this question.

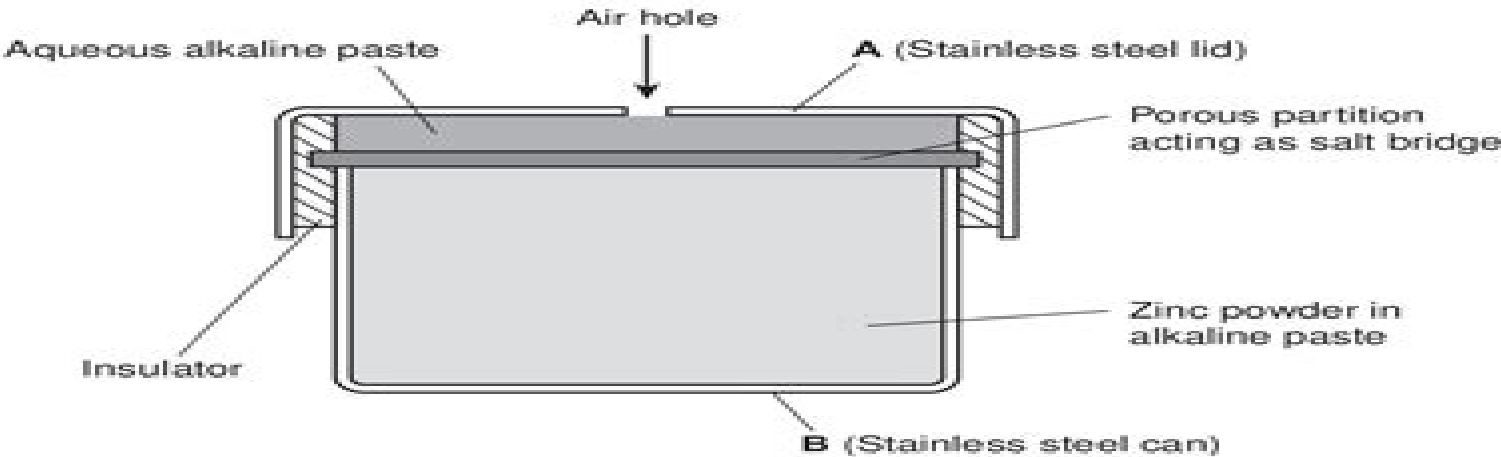
Table 3

	$E^{\ominus} / \text{V}$
$\text{ZnO(s)} + \text{H}_2\text{O(l)} + 2\text{e}^{-} \longrightarrow \text{Zn(s)} + 2\text{OH}^{-}(\text{aq})$	-1.25
$\text{Fe}^{2+}(\text{aq}) + 2\text{e}^{-} \longrightarrow \text{Fe(s)}$	-0.44
$\text{O}_2(\text{g}) + 2\text{H}_2\text{O(l)} + 4\text{e}^{-} \longrightarrow 4\text{OH}^{-}(\text{aq})$	+0.40
$2\text{HOCl(aq)} + 2\text{H}^{+}(\text{aq}) + 2\text{e}^{-} \longrightarrow \text{Cl}_2(\text{g}) + 2\text{H}_2\text{O(l)}$	+1.64

The half-equations from Table 3 that involve zinc and oxygen are simplified versions of those that occur in hearing aid cells.

A simplified diagram of a hearing aid cell is shown in Figure 1.

Figure 1



5 (f) (i) Use data from Table 3 to calculate the e.m.f. of this cell.

[1 mark]

.....

.....

Answer =

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