Guestion	Answer	Marks	Guidance	
1 5 1	Group 7	11		
1.6.1	Group 6	- 1	1	
TAR	Group 1	1		
GCSE Che	mintry only	500		
1 a iv	between Groups 2 and 3			
161	transition elements	11		
1 6 8	Group 1 / alkali metals	11		
1 D III	Group 0 / noble gases	1		
t bily	Group 7 / halopens	1 1 1		
2 a .	7 electrons as Group 7	1.0		
2 b	solid	1		
101	TORIC	100		
2 (0.18)	white	1		
2 (2 10)	NaAl	100		
2 0 IV	sodium + astatine → sodium astatide 2Na + At, → 2NaAt	1		
ZCV.	yes, chlorine more reactive than astaline	1		

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C2 Summary questions

Question	Answer	Marks	Guidance
2-6	F. Cl. Br. I. Al. The smaller the halogen atom, the more readily it accepts/ gains entra electron into its outermost shell, becomes halide ion with single negative charge, the stronger the electrostatic force of attraction between nucleus and extra electron entening smaller atom's outermost shell as electron closer to nuclear charge, and it becomes less shielded from nuclear charge than larger atoms with more inner shells of electrons. These two factors outweigh fact that nuclear charge is more positive going down Ciroup 7.	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
3 4	soft good electrical conductivity low meting point	1	
3.61	1+	11	
361	Rubidium compound Chemical formula nubidium iodide Rbi nubidium fluoride RbF nubidium fluoride RbOH	3	
3 6 1	29b(s) + 2H ₂ O(t) → 2RbOH(aq) + H ₂ (g)	3	
368	2FB(s) + CL(g) 2FB(Cl(s)	12.	

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these British behemoths was the Albert Ballin designed trio of Imperator Vaterland and Bismarck Through misfortune or war two of these vessels would sink but the others led useful lives with Aquitania surviving two world wars before being scrapped Designed to be the absolute engineering achievements of their time these nine vessels dominated the Atlantic J Kent Layton tells the story of the Edwardian Superliners in this fabulously illustrated volume showcasing many images previously unpublished and never before seen Rarely can one describe a book as definitive but this volume truly deserves the accolade **AQA GCSE Chemistry** ,2011

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