AQA Chemistry GCSE Student Book answers

C5 Summary questions

Question	Answer	Marks	Guidance
1 a	fizzes readily, X gets smaller and eventually dissolves completely X * sulfuric acid X sulfate * hydrogen X(s) * H _s SO _s (aq) XSO _s (aq) * H _s (q)	1 1 1 2	
1 b	Magnesium reads very slowly with water, calcium reads vigorously with water. We do not know how much less reactive than calcium and more readilive than magnesium the element X is and so cannot predict whether the reaction will be slow like magnesium or more vigorous like calcium.	1	
1ci	the more reactive metal X displaces less reactive copper so solution loses its blue colour, X dissolves into solution and copper metal deposited as a brown solid X(s) = Cu(NO ₁) ₂ (eq) X(NO ₁) ₂ (eq) + Cu(s)	1 1 1	
O tel	$X(s) + Cu^{2\gamma}(aq) \rightarrow X^{0\gamma}(aq) + Cu(s)$	3	
0 1 cii	$X(s) \rightarrow X^{0}(aq) + 2e -$ X atoms oxidised as lose 2 electrons $Cu^{0}(aq) + 2e^{-} \rightarrow Cu(s)$ Cu^{0} ions reduced as gain 2 electrons.	1 1 1 1	
1 d i	just above copper	1	
1 d ii	no change / no reaction as Y less reactive than magnesium so cannot displace it from solution of one of its compounds	1 1	
2.61	react	1	
241	no reaction	1 .	

Aqa Science Chemistry Summary Question Answer

Patrick Fullick

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- 1. Understanding the eBook Aga Science Chemistry Summary Question Answer
 - The Rise of Digital Reading Aga Science Chemistry Summary Question Answer
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Aqa Science Chemistry Summary Question Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Aga Science Chemistry Summary Question Answer
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Aga Science Chemistry Summary Question Answer
 - Personalized Recommendations
 - Aga Science Chemistry Summary Question Answer User Reviews and Ratings
 - Aga Science Chemistry Summary Question Answer and Bestseller Lists

- 5. Accessing Aqa Science Chemistry Summary Question Answer Free and Paid eBooks
 - Aga Science Chemistry Summary Question Answer Public Domain eBooks
 - Aqa Science Chemistry Summary Question Answer eBook Subscription Services
 - Aga Science Chemistry Summary Question Answer Budget-Friendly Options
- 6. Navigating Aqa Science Chemistry Summary Question Answer eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Aga Science Chemistry Summary Question Answer Compatibility with Devices
 - Aqa Science Chemistry Summary Question Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Aqa Science Chemistry Summary Question Answer
 - Highlighting and Note-Taking Aga Science Chemistry Summary Question Answer
 - Interactive Elements Aga Science Chemistry Summary Question Answer
- 8. Staying Engaged with Aqa Science Chemistry Summary Question Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Aqa Science Chemistry Summary Question Answer
- 9. Balancing eBooks and Physical Books Aqa Science Chemistry Summary Question Answer
 - ∘ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Aga Science Chemistry Summary Question Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Aga Science Chemistry Summary Question Answer
 - Setting Reading Goals Aga Science Chemistry Summary Question Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Aga Science Chemistry Summary Question Answer
 - Fact-Checking eBook Content of Aqa Science Chemistry Summary Question Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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