

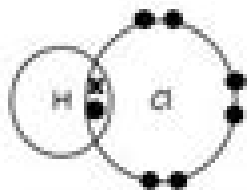
C3 - STRUCTURE AND BONDING 3.5 Covalent bond

Aiming for Grade 4

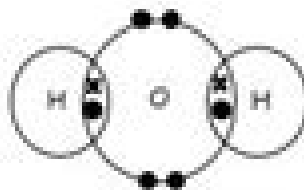
Ionic bonds form between **metals** and **non-metals**. Covalent bonds form between **non-metals**. Covalent bonds form when atoms **share** pairs of **electrons**. These bonds are very **strong**.

Share, Metals, strong, non-metals, electrons
You can use the words more than once.

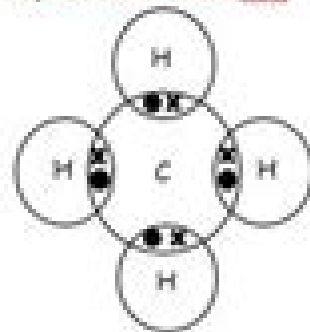
Name these small molecules and state their formulae.



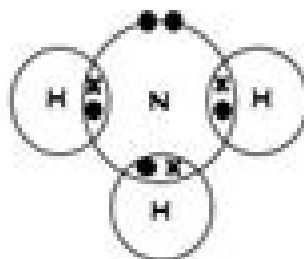
Hydrochloric Acid (HCl)



Water (H_2O)



Methane (CH_4)



Ammonia (NH_3)

Assessment

Aiming for Grade 6

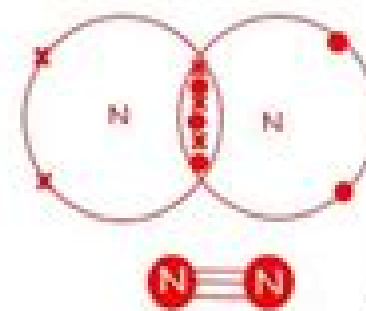
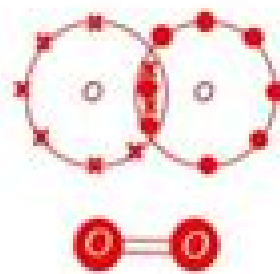
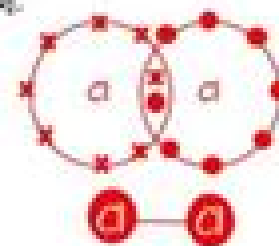
Describe how Hydrogen and Chlorine form a covalent bond to form Hydrochloric acid.

To gain full outer shells Hydrogen needs to gain an electron and Chlorine needs to lose one. Hydrogen shares its outer shell electron with Chlorine and Chlorine shares one of its electrons.

What is a double covalent bond?

Two shared pairs of electrons

Draw dot and cross diagrams and ball and stick diagrams for H_2 , Cl_2 , O_2 , and N_2 .



Assessment

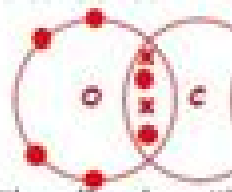
Aiming

Draw ball and stick or the following covalent

Three hydrogen atoms
one phosphorus (phos)



Carbon dioxide



Ethane (2 carbon with hydrogen)



How do double and triple bonds affect properties?

A double or triple bond the bond making the me

Assessment
Grade 6

1. I can draw dot and

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Iseb Iseb



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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Aqa Chemistry C3 May 2013 Mark Scheme**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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