

AQA Physics GCSE Student Progress

Unit 4.3 - Particle Model of Matter



4.3.1. Changes of State and the Particle Model

4.3.1.1. Density of Materials

a	I can recall and apply the following equation: density = mass / volume $\rho = m / V$ density, ρ , in kilograms per metre cubed, kg/m^3 mass, m , in kilograms, kg volume, V , in metres cubed, m^3			
b	I know that the particle model can be used to explain: • the different states of matter; • differences in density.			
c	I can recognise / draw simple diagrams to model the difference between solids, liquids and gases.			
d	I can explain the differences in density between the different states of matter in terms of the arrangement of atoms or molecules.			

4.3.1.2. Changes of State

a	I can describe how, when substances change state (melt, freeze, boil, evaporate, condense or sublimate), mass is conserved.			
b	I know that changes of state are physical changes which differ from chemical changes because the material recovers its original properties if the change is reversed.			

4.3.2. Internal Energy and Energy Transfers

4.3.2.1. Internal Energy

a	I know that energy is stored inside a system by the particles (atoms and molecules) that make up the system. This is called internal energy.			
b	I know that internal energy is the total kinetic energy and potential energy of all the particles (atoms and molecules) that make up a system.			
c	I know that heating changes the energy stored within the system by increasing the energy of the particles that make up the system. This either raises the temperature of the system or produces a change of state.			

Aqa Physics Unit 3 Isa 2013

Michael Lacewing



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Parliamentary Debates (Hansard). Great Britain. Parliament. House of Commons, 2012

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