
AQA A LEVEL PHYSICS PAPER 1

1. What is meant by work function (2): Minimum energy required by an electron to escape from (1) a metal surface (1)

2. What is meant by ionisation energy? (2): Minimum energy required to remove an electron from an atom (1) from the ground state (1)

3. Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4): Energy of a photon is fixed/constant (1)

Energy is required for electrons to overcome work function ϕ (1)

Electrons deeper into/below the surface of the metal require more energy to overcome work function and therefore have less KE (1)

Maximum KE = Energy of photon - work function ϕ (1)

4. Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons?: Diffraction pattern maxima/minima (bright and dark areas) (1) Diffraction is a property of waves- proves wave-behaviour of electrons (1)

5. What is meant by the ground state?: The lowest energy state of an atom

6. Explain the difference between excitation and ionisation (2): An electron receives a discrete amount of energy for BOTH (1)

Excitation promotes an electron to a higher energy level

Ionisation is when an electron receives enough energy to leave the atom

7. Explain why only photons of certain frequencies cause excitation (4): discrete energy levels (1)

Electrons need to absorb a DISCRETE amount of energy to move to a higher energy level (1)

Aqa Physics Empa Questions 2015

Rachel Sandford



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