C2 Moles and Calculations Exam Questions

 Iron is an essential part of the human diet. Iron(II) sulfate is sometimes added to white bread flour to provide some of the iron in a person's diet.



(a) The formula of iron(II) sulfate is FeSO,

Calculate the relative formula mass (M,) of FeSO4

Relative atomic masses: O = 16; S = 32; Fe = 56.

(b) What is the mass of one mole of iron(II) sulfate? Remember to give the unit.

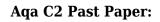
(1 mark)

(c) What mass of iron(II) sulfate would be needed to provide 28 grams of iron? Remember to give the unit.

(1 mark)

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Liying Dong



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