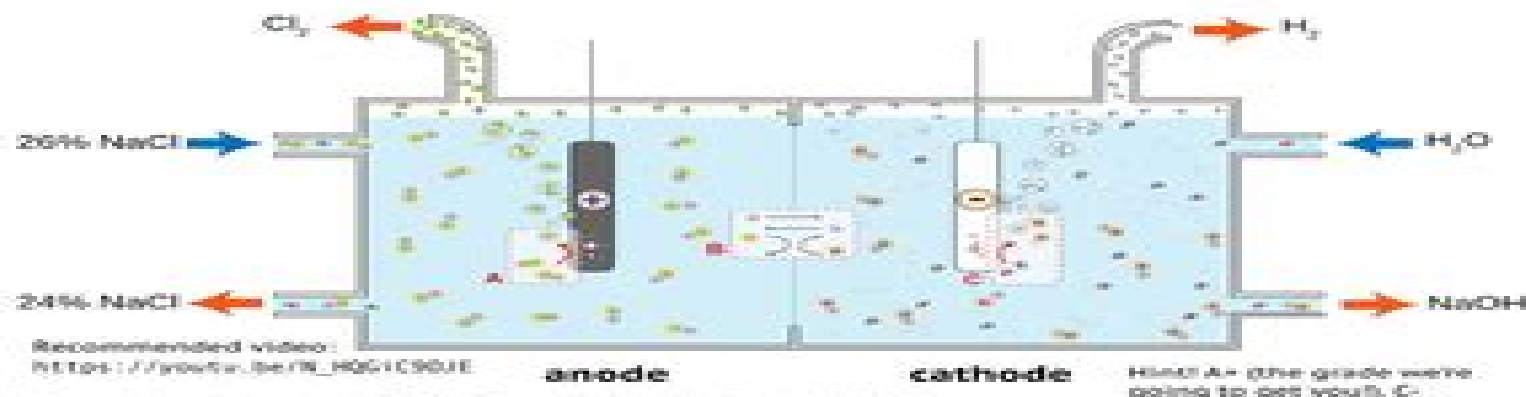


AQA electrolysis of solutions equations



1. Chemistry has a rule OIL RIG: explain what it stands for?

2. When you do electrolysis, you have an anode-what's that?

3. What's the cathode?

4. At the cathode, you normally get hydrogen, unless the solution has a metal ion in it that's below hydrogen in the reactivity series.
 Try to name two metals below hydrogen in the reactivity series.

5. What's the rule for what you get at the anode?

6. So what do you get when you electrolyse sodium chloride solution?
 (Also called salt water or brine.) You can use the diagram.

Cathode

Anode

7. And what's left behind in the water?
 (Hint: look at the diagram again)

Suggest what pH the solution is going to be at the end?

8. What reaction breaks up water?

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Yicheng Fang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Yicheng Fang.

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