

# 1999 Advanced Placement Chemistry Exam

## Section I: Multiple Choice

Time — 1 hour and 30 minutes

NO CALCULATORS MAY BE USED WITH SECTION I.

Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions specified unless otherwise noted.

$T$  = temperature  
 $P$  = pressure  
 $V$  = volume  
 $S$  = entropy  
 $H$  = enthalpy  
 $G$  = free energy  
 $R$  = molar gas constant  
 $n$  = number of moles  
 $M$  = molar

$m$  = molal  
 $L, mL$  = liter(s), milliliter(s)  
 $g$  = gram(s)  
 $nm$  = nanometer(s)  
 $atm$  = atmosphere(s)  
 $J, kJ$  = joule(s), kilojoule(s)  
 $V$  = volt(s)  
 $mol$  = mole(s)

### Part A

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then fill in the corresponding oval on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1–4 refer to the following types of energy.

- (A) Activation energy
- (B) Free energy
- (C) Ionization energy
- (D) Kinetic energy
- (E) Lattice energy

1. The energy required to convert a ground-state atom in the gas phase to a gaseous positive ion
2. The energy change that occurs in the conversion of an ionic solid to widely separated gaseous ions
3. The energy in a chemical or physical change that is available to do useful work
4. The energy required to form the transition state in a chemical reaction

Questions 5–8 refer to atoms for which the occupied atomic orbitals are shown below.

- (A)  $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}}$
- (B)  $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}}$
- (C)  $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}} 2p \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$
- (D)  $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}} 2p \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$
- (E)  $[Ar] 4s \underline{\hspace{1cm}} 3d \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$

5. Represents an atom that is chemically unreactive
6. Represents an atom in an excited state
7. Represents an atom that has four valence electrons
8. Represents an atom of a transition metal

Questions 9–12 refer to aqueous solutions containing 1:1 mole ratios of the following pairs of substances. Assume all concentrations are 1  $M$ .

# 2008 Ap Chemistry Exam Multiple Choice Answers

**James H. McMillan**



## **2008 Ap Chemistry Exam Multiple Choice Answers:**

*5 Steps to a 5 AP Chemistry, 2008-2009 Edition* John T. Moore, Richard H. Langley, 2008-01-04 A PERFECT PLAN FOR THE PERFECT SCORE We want you to succeed on your AP exam That s why we ve created this 5 step plan to help you study more effectively use your preparation time wisely and get your best score This easy to follow guide offers you a complete review of your AP course strategies to give you the edge on test day and plenty of practice with AP style test questions You ll sharpen your subject knowledge strengthen your thinking skills and build your test taking confidence with Full length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules so you can pick the one that meets your needs The 5 Step Plan helps you get the most out of your study time Step 1 Set Up Your Study Program Step 2 Determine Your Readiness Step 3 Develop the Strategies Step 4 Review the Knowledge Step 5 Build Your Confidence **Guide to College Majors 2008** Princeton Review (Firm), 2008-02-05 Fully revised and updated for 2008 this resource includes the most current information available on more than 350 of the most popular majors Each profile includes an overview of the major career options and earning potential sample college curricula and more **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP

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and validating selected response test items Like its predecessor the content of this new book is based on both an extensive review of the literature and on its author s long experience in the testing field It is very timely in this era of burgeoning testing programs especially when these items are delivered in a computer based environment Key features include Comprehensive and Flexible No other book so thoroughly covers the field of test item development and its various applications Focus on Validity Validity the most important consideration in testing is stressed throughout and is based on the Standards for Educational and Psychological Testing currently under revision by AERA APA and NCME Illustrative Examples The book presents various selected and constructed response formats and uses many examples to illustrate correct and incorrect ways of writing items Strategies for training item writers and developing large numbers of items using algorithms and other item generating methods are also presented Based on Theory and Research A comprehensive review and synthesis of existing research runs throughout the book and complements the expertise of its authors *The Publishers Weekly* ,2008

**Executive Thinking** Kevin Baize OD,Morris Graham PhD,2011-02-18 Life is about how much we think Thinking is about how much mental capacity we possess Capacity in addition to our abilities and conscientiousness is about how much we can process combinations of verbal height quantitative width and spatial depth with decisiveness direction and speed No matter where we go or what we do as executives we take our thinking with us That may spoil everything because to a great extent we do and accomplish what we think about Our thoughts mold our aspirations attitudes and accomplishments during our life In other words our careers and lives are influenced more by the power of our thoughts than anything else The bad news is that most of us never fully use our mental capacities and never achieve our potential The good news is that neural technologies are now available to transform our thinking into the higher realms of brilliance Developing the spatial capacity to think higher wider and deeper means breaking away from the effects of years of flat thinking or educational backgrounds that stifles creative innovative potential Expand your mental agility through a development of higher order processes and discover a whole new world mentally in Executive Thinking **Guide to College Visits** Princeton Review,2007-04-10 A practical handbook covers everything needed to plan a successful visit to campuses across the country with information on nearby accommodations for every budget the best and worst times to visit itineraries and campus tours **Excel With Objective Questions In Chemistry** Prof. S. K. Khanna,Dr. N. K. Verma,Dr. B. Kapila,2006

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