Internship Report

Description of the internship

2.1 The organization Faunagua

Faunagua is a non-governmental organization (NGO) founded in 2004. It works on the development implementation of projects, which contribute to local development, sustainable use and conservation the aquatic ecosystems and resources in Bolivia. The organization facilitates the collaboration betwee local communities, investigators, national authorities and private institutions. In its vision the aspect c local people is very important. Therefore, in all projects local people are taken into account. By respect customs and traditional knowledge and without danger for the environment, Faunagua tries to improthe quality of life of local populations (indigenous communities, rural people and fishermen).

An overview of the projects of Faunagua:

- The development of strategies for sustainable exploitation, management and conservation of hydrobiological resources and fishery;
- Studies on the biology and ecology of fauna (Pteronura brasiliensis, Inia boliviensis)
- Identification of threats for aquatic fauna and fishery
- Development and implementation of management plans for aquatic fauna (Several fish speci-Turtels: Podocnemis expansa and P. unifilis; Caiman: Caiman yacare; Giant otter: Pteronura brasiliensis; River dolphin: Inia boliviensis);
- Monitoring: fauna and aquatic habitats
- Establishment and monitoring of alternative production networks of handicrafts of crocodile fish skins
- Development of cartographic information
- Strengthening of responsible tourism
- Studies on the possible impact of hydroelectrical dam constructions

For NGO's the availability of resources is one of the main factors that determines the scope and durati of projects. To be able to carry out projects of several months, but also projects of several years, sufficient external financial support is needed. Each financier has its own criterions for applying projects.

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effective for student learning particularly for the development of holistic competencies i e 21st century skills soft skills transferable skills considered essential for individuals to succeed in the increasingly global and technology infused 21st century society Universities around the world are now actively organising experiential learning activities or programmes for students to gain enriching and diversified learning experiences however the assessment of these programmes tends to be limited unclear and contested Assessment plays a central role in education policies and students approach to learning But do educators know how to assess less traditional learning such as service learning entrepreneurship cross discipline or cross cultural projects internships and student exchanges While the current assessment landscape is replete with assessments that measure knowledge of core content areas such as mathematics law languages science and social studies there is a lack of assessments and research that focus on holistic competencies How do we assess students ability to think critically problem solve adapt self manage and collaborate Central to the discussion in this book is the reason students are assessed and how they should be assessed to bring out their best learning outcomes Offering a collection of best assessment practice employed by teachers around the world this volume brings together both theoretical and empirical research that underpins assessment and perceptions of different stakeholders understanding of assessment in experiential learning from students teachers and policymakers The idea of assessment literacy also plays an important role in experiential learning for example reflection is often used in assessing students in experiential learning but how reflection literate are educators are they aware of the ethical dilemmas that arise in assessing students These questions are discussed in detail The volume also introduces a quality assurance programme to recognise student development within experiential learning programmes The book will be particularly informative to academic developers teachers students and community partners who struggle with the development and assessment for experiential learning those who plan to apply for funding in experiential learning and policymakers and senior managements seeking evidence and advice on fine tuning curricular assessment designs and quality assurance The Open Access version of this book available at www taylorfrancis com has been made available under a Creative Commons Attribution Non Commercial No Derivatives 4 0 license Lawyer Hiring & Training Report, 1985

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web nov 14 2023 show all resonance structures possible use formal charges to pick the best form if necessary the following molecule has two lewis dot formulas write one that uses 8 electrons about the central atom and one that expands the natat 9 lewis structures and molecular shapes experiment - Sep 18 2023

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web science chemistry chemistry questions and answers ch 201 lab exercise molecular modeling purpose to practice lewis dot structures and associate them with models to use molecular modeling software and compare the results to static models discussion in this exercise you will get practice drawing electron dot structures for organic compounds lewis dot structure and vsepr theory lab practical 3 lewis dot - Nov 08 2022

web 2019 0501140012 american native medicine 1 2019 0612171451 orpic company 1 biome viewer student ws copy of lab 1 effect environment se the perodic properties of the main group to understand theory behind lewis structure and valence shell electron pair repulsion theory to experience molecule s geometrical shapes by the model that has lewis structure quiz proprofs quiz - Feb 28 2022

web mar 27 2023 a the lewis structure for water is represented by the given option a in this structure the oxygen atom is in the center and is bonded to two hydrogen atoms the oxygen atom has two lone pairs of electrons this structure follows the octet rule where each atom has a full valence shell with eight electrons

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web introduction the physical and chemical behavior of compounds depends on the types of bonds that exist within the molecule and on the overall shape of the molecules the branch of chemistry that investigates how molecules are put together is

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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl 9 3 drawing lewis structures chemistry libretexts - Jun 03 2022

web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web science chemistry chemistry questions and answers molecular modeling purpose in this lab you will employ your skill of drawing lewis dot structures to help you build models of covalently bonded compounds

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web molecular models and 3d printing activity lewis dot structures and molecule geometries worksheet answer key 6 10 co 3 2 total number of valence electrons 24 cad engineered 3d sketch model show dipole arrows lewis structure is there a polar bond in this molecule yes or no vsepr shape name trigonal planar bond angles 120 lewis structure in chemistry sciencedaily - May 02 2022

web jun 8 2022 lewis structure in chemistry lewis structures also called electron dot structures or electron dot diagrams are diagrams that show the bonding between atoms of a molecule and the lone pairs of

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry