

2024 ACADEMIC CALENDAR

JANUARY

M	T	W	T	F	S	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

FEBRUARY

M	T	W	T	F	S	S
			1	2	3	4
5	6	7	8	9	10	11
12	13	14	15	16	17	18
19	20	21	22	23	24	25
26	27	28	29			

MARCH

M	T	W	T	F	S	S
				1	2	3
4	5	6	7	8	9	10
11	12	13	14	15	16	17
18	19	20	21	22	23	24
25	26	27	28	29	30	31

APRIL

M	T	W	T	F	S	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

MAY

M	T	W	T	F	S	S
		1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30	31		

JUNE

M	T	W	T	F	S	S
					1	2
3	4	5	6	7	8	9
10	11	12	13	14	15	16
17	18	19	20	21	22	23
24	25	26	27	28	29	30

JULY

M	T	W	T	F	S	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

AUGUST

M	T	W	T	F	S	S
			1	2	3	4
5	6	7	8	9	10	11
12	13	14	15	16	17	18
19	20	21	22	23	24	25
26	27	28	29	30	31	

SEPTEMBER

M	T	W	T	F	S	S
						1
2	3	4	5	6	7	8
9	10	11	12	13	14	15
16	17	18	19	20	21	22
23	24	25	26	27	28	29

OCTOBER

M	T	W	T	F	S	S
	1	2	3	4	5	6
7	8	9	10	11	12	13
14	15	16	17	18	19	20
21	22	23	24	25	26	27
28	29	30	31			

NOVEMBER

M	T	W	T	F	S	S
				1	2	3
4	5	6	7	8	9	10
11	12	13	14	15	16	17
18	19	20	21	22	23	24
25	26	27	28	29	30	

DECEMBER

M	T	W	T	F	S	S
						1
2	3	4	5	6	7	8
9	10	11	12	13	14	15
16	17	18	19	20	21	22
23	24	25	26	27	28	29

PUBLIC HOLIDAYS

- NEW YEAR'S DAY
MONDAY 1 JANUARY
- AUSTRALIA DAY
26 JANUARY
- GOOD FRIDAY
29 MARCH
- EASTER SATURDAY
30 MARCH
- EASTER SUNDAY
31 MARCH
- ANZAC DAY
25 APRIL
- KING'S BIRTHDAY
10 JUNE
- LABOUR DAY
7 OCTOBER
- CHRISTMAS DAY
25 DECEMBER
- BOXING DAY
26 DECEMBER

CALENDAR KEY

- INTAKE DATES (DI&ADI&ADT)
- PUBLIC HOLIDAY
- TERM BREAK
- PRE-TEST TRAINING (CPL, CL, CT)

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web arrhenius bases an arrhenius base is defined as any species that increases the concentration of hydroxide ions oh⁻ in aqueous solution an example of an arrhenius base is the highly soluble sodium hydroxide naoh sodium hydroxide dissociates in water as follows naoh(aq) → na⁺(aq) + oh⁻(aq) in water sodium hydroxide

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web key points a brønsted lowry acid is any species that is capable of donating a proton H^+ a brønsted lowry base is any species that is capable of accepting a proton which requires a lone pair of electrons to bond to the H^+ water is amphoteric which means it can act as both a brønsted lowry acid and a brønsted lowry base

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