

Answer Sheet

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Answer Of The Test Night

Tao Wei



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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$ because the mass number is the same on both sides of the equation x must have a mass number of 0

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remember that the two ${}^{94}_{40}\text{Ne}$ neutrons each contribute

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