#### Chemistry Final Exam Review

#### Multiple Choice

S. Samericki	Albert Bertham	South Bellevier	with the said of t	Same States	Commenced.	Same Steel	n i profession Participano a come del	or answers ti	har amount form
SERVICE PROPERTY.	CORP. CHILDREN	with water	CHICARCIE SE	AND RESIDENCE OF THE PARTY OF T	Carlo Street Shakes	STREET, STREET,	CONTRACTOR CONTRACTOR	WALL BRAINS HOLDER OF THE	one: of energy processor.

1.	The study of chemicals that, in general, do not contain carbon is traditionally called what type of chemistry									
	a, bio	e.	physical							
	b. inorganic	d.	analytical							
2.	Which of the following is NOT an example	of mat	ter?							
	a, air	400	smoke							
	b. heat	d.	water vapor							
3.										
	a. takes up more space	C.	contains different kinds of matter							
	b, contains more matter	d.	has a definite composition							
4.	An example of an extensive property of matter is									
	a. temperature	(c)	mass							
	b. pressure	d.	hardness							
5.	All of the following are physical properties of matter EXCEPT									
	a, mass	et.	melting point							
	b. color	d.	ability to rust							
6.	Which state of matter has a definite volume and takes the shape of its container?									
	a, solid	C.	gas							
	b. liquid	d.	both b and c							
7.	Which of the following CANNOT be classified as a substance?									
	a. table salt	C.	nitrogen							
	b. air	d.	gold							
8.	Which of the following is a heterogeneous mixture?									
		C.								
	b. salt water	d.	soil							
9.	Which of the following is a heterogeneous mixture?									
	a. vinegar in water	Ch.	oil and vinegar							
	b. milk	d.	air							
10.	Separating a solid from a liquid by evaporat	ing the	liquid is called							
	a. filtration	C.	solution							
	b. condensation	d.	distillation							
11.	A substance that can be separated into two or more substances only by a chemical change is a(n)									
		C.								
	b. element	d.	compound							
12.										
	n. fe	C.	Fe							
	b. FE	d.	lr .							
	4. 5. 6. 7. 8. 9.	<ol> <li>a. bio b. inorganic</li> <li>Which of the following is NOT an example a. air b. heat</li> <li>A golf ball has more mass than a tennis ball a. takes up more space b. contains more matter</li> <li>An example of an extensive property of mat a. temperature b. pressure</li> <li>All of the following are physical properties a. mass b. color</li> <li>Which state of matter has a definite volume a. solid b. liquid</li> <li>Which of the following CANNOT be classif a. table salt b. air</li> <li>Which of the following is a heterogeneous ma. air b. salt water</li> <li>Which of the following is a heterogeneous ma. air b. salt water</li> <li>Which of the following is a heterogeneous ma. vinegar in water b. milk</li> <li>Separating a solid from a liquid by evaporat a. filtration b. condensation</li> <li>A substance that can be separated into two ca. solution b. element</li> <li>The chemical symbol for iron is</li> </ol>	a. bio b. inorganic c. d.  2. Which of the following is NOT an example of mat a. air b. heat c. d.  3. A golf ball has more mass than a tennis ball becaus a. takes up more space b. contains more matter d.  4. An example of an extensive property of matter is a. temperature b. pressure d.  5. All of the following are physical properties of mat a. mass b. color d.  6. Which state of matter has a definite volume and ta a. solid c. b. liquid d.  7. Which of the following CANNOT be classified as a. table salt b. air d.  8. Which of the following is a heterogeneous mixture a. air b. salt water d.  9. Which of the following is a heterogeneous mixture a. air c. b. milk d.  10. Separating a solid from a liquid by evaporating the a. filtration b. condensation d.  11. A substance that can be separated into two or more a. solution b. clement d.  12. The chemical symbol for iron is a. fe							

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Jodi Weiss, Princeton Review (Firm)

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