#### Mixed Gas Laws Worksheet - Solutions

 How many moles of gas occupy 98 L at a pressure of 2.8 atmospheres and a temperature of 292 K?

2) If 5.0 moles of O<sub>2</sub> and 3.0 moles of N<sub>2</sub> are placed in a 30.0 L tank at a temperature of 25° C, what will the pressure of the resulting mixture of gases be? 25° C = 298 K

$$P_{Tot} = P_{02} + P_{N2} = 4.1 \text{ atm} + 2.4 \text{ atm} = 6.5 \text{ atm}$$

A balloon is filled with 35.0 L of helium in the morning when the temperature is 20.0° C.
 By noon the temperature has risen to 45.0° C. What is the new volume of the balloon?

$$T_1 = 20.0^{\circ} C = 293 \text{ K, } V_1 = 35.0 \text{ L, } T_2 = 45.0^{\circ} C = 318 \text{ K, } V_2 = ?$$

$$V_2 = \frac{V_1 \cdot T_2}{T_1} = \frac{(35.0 \text{ L})(318 \text{ K})}{(293 \text{ K})} = 38.0 \text{ L}$$

4) A 35 L tank of oxygen is at 315 K with an internal pressure of 190 atmospheres. How many moles of gas does the tank contain?

5) A balloon that can hold 85 L of air is inflated with 3.5 moles of gas at a pressure of 1.0 atmosphere. What is the temperature in <sup>6</sup>C of the balloon?

$$T = \frac{PV}{nR} = \frac{(1 \text{ atm})(85 \text{ L})}{(3.5 \text{ mol})(0.0821 \text{ Latm/mol K})} = 296 \text{ K} = 23^{\circ} \text{ C}$$

6) CaCO<sub>2</sub> decomposes at 1200° C to form CO<sub>2</sub> gas and CaO. If 25 L of CO<sub>2</sub> are collected at 1200° C, what will the volume of this gas be after it cools to 25° C?

$$T_1 = 1200^{\circ} C = 1473 \text{ K}, V_1 = 25 \text{ L}, T_2 = 25^{\circ} C = 298 \text{ K}, V_2 = ?$$

$$V_2 = \frac{V_1 T_2}{T_1} = \frac{(25 L)(298 K)}{(1473 K)} = 5.1 L$$

7) A helium balloon with an internal pressure of 1.00 atm and a volume of 4.50 L at 20.0° C is released. What volume will the balloon occupy at an altitude where the pressure is 0.600 atm and the temperature is -20.0° C?

$$P_1=1.00$$
 atm,  $V_1=4.50$  L,  $T_1=20.0^{\circ}$  C = 293 K,  $P_2=0.600$  atm,  $V_2=?$  ,  $T_2=-20.0^{\circ}$  C = 253 K

$$V_2 = P_1 V_1 T_2 = (1.00 \text{ atm})(4.50 \text{ L})(253 \text{ K}) = 6.48 \text{ L}$$
  
 $T_1 P_2$  (293 K)(0.600 atm)

8) There are 135 L of gas in a container at a temperature of 260° C. If the gas was cooled until the volume decreased to 75 L, what would the temperature of the gas be?

$$T_1 = 260^{\circ} C = 533 \text{ K}, V_1 = 135 \text{ L}, T_2 = 7, V_2 = 75 \text{ L}$$

$$T_2 = \frac{V_2 T_1}{V_1} = \frac{(75 L)(533 K)}{(135 L)} = 296 K = 23^{\circ} C$$

# **Answer Sheet To Mixed Gas Laws**

Christopher R. Hren, John T. Moore, Peter J. Mikulecky

#### **Answer Sheet To Mixed Gas Laws:**

U.S. Navy Diving Manual: Mixed-gas diving ,1991 U.S. Navy Diving Manual United States. Naval Sea Systems Chemistry All-in-One For Dummies (+ Chapter Ouizzes Online) Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you ll need to pass your chemistry class with flying colors You ll find tons of practical examples and practice problems and you ll get access to an online guiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter guizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or Method and Appraisal in the Physical Sciences Colin want to cram extra hard to ace that chem test Howson, 1976-09-02 This is a volume of studies on the problems of theory appraisal in the physical sciences **GATE** Environment Science & Engineering [ES] Solved Previous Year Question Paper II Latest BookII 2021 to 2025 II **PYO II With Detail Solution**, 2025-06-24 Prepare thoroughly for the GATE Environmental Science Engineering ES examination with this comprehensive collection of fully solved previous year question papers 2021 2025 Designed specifically for GATE aspirants this book provides detailed step by step solutions to each question helping students grasp key concepts identify trends and master the GATE ES syllabus with confidence Key features of the book Complete Coverage 2021 2025 Detailed Solutions Topic wise Insights Updated as per Latest Pattern Ideal for Practice and Revision **CliffsNotes** Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level 10000 MCQ - General Studies Previous Paper Based Question

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