

8. Buffer, passing question 10, is shown within the Earth's crust. Buffer is composed of the components of those of existing rocks.

A sample of Buffer from a sample was analyzed to give the following composition of samples.

Sample	Composition (%)
100	100.00
100	10.00
100	10.00

- (a) Define the term relative atomic mass.

The relative atomic mass of an element is the mass of one atom of the element divided by the mass of one atom of ^{12}C .

- (b) Calculate the relative atomic mass of the sample of buffer.

Give your answer to three decimal places.

$$\left(\frac{100 \times 100}{100} \right) + \left(\frac{10 \times 10}{100} \right) + \left(\frac{10 \times 10}{100} \right)$$

100.00

Answer = 100.00

- (c) Some studies, on early fossil, indicate that specimens were made up of long particles which could be used to form the structure, structure known as the structure of subatomic particles in atoms and in ions.

Complete the table to show the number of subatomic particles in the ^{100}B atom and $^{100}\text{B}^{+}$ ion.

	protons	neutrons	electrons
^{100}B	10	10	10
$^{100}\text{B}^{+}$	10	10	9



Answer To F321 Chemistry May Series

Mike Smith



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