d potassium and bromine
 e sodium and bromine

23

3

Complete the word equations for the reactions of the Group 7 elements with metals. iron + ...Lh loth .... ..... → iron chloride iron + bromine - ... Itak bromide sodium + ...... → sodium iodide potassium + bromine - pa tessisson bramiele lithium + ..... (A.la.r.in.S....... → lithium chloride sodium + chloring - sodium ( 4 km de. 2 Make up four sentences using the phrases in the table. Each sentence must include a phrase from each column. between the two atoms of a chlorine molecule. weak forces between a chlorine molecule and its neighbours. There are strong forces between the two atoms of an iodine molecule. between an iodine molecule and its neighbours. Have one week forces between he that along of a charle whenhe Here are stong loses between a chlame rolefule and its very los lever letteren the two stors do an lective moterte Ruces letween on today mobile and its witch In 3 For the sentences A-H below: Write 1 next to the sentences that are true for Group 1. Write 7 next to the sentences that are true for Group 7. Write B next to the sentences that are true for both Groups 1 and 7. Going down this group, boiling point increases. Going up this group, the reactions of the elements get more vigorous. Going down this group, the number of protons in atoms of the elements increases. Atoms of the elements in this group form molecules made up of two atoms. Going up this group, the colours of the elements get lighter. The elements at the top of this group are gases. G. The elements in this group conduct electricity. H. All the atoms of the elements in this group have the same number of electrons in the outermost shell. Write M next to the pair of elements that you predict will react most vigorously. Write L next to the pair of elements that you predict will react least vigorously. a sodium and iodine b lithium and lodine c potassium and fluorine

# **Answers For A Ls Chemistry**

**Charles E. Mortimer** 

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