

20.1

DESCRIBING ACIDS AND BASES

SECTION REVIEW

Objectives

- List the properties of acids and bases.
- Name an acid or base when given the formula.

Key Terms

- acid
- base

Part A Completion

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

- Aqueous solutions of _____ 1 _____ are electrolytes. Acids react with many _____ 2 _____ to produce _____ 3 _____ gas. _____ 4 _____ are compounds that react with acids to form _____ 5 _____ and a salt.
- Acids produce _____ 6 _____ ions when dissolved in water. There is a standard system for naming acids. For example, HBr is called _____ 7 _____ acid, while HNO_3 is called _____ 8 _____ acid.
- Bases produce _____ 9 _____ ions when dissolved in water. Some compounds that are _____ 10 _____ are named in the same way as other ionic compounds.
1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

Part B True-False

Circle the word of these classes as true or false: acid, base, OH^- ion, H^+ ion, HCl , HNO_3 , H_2SO_4 , H_2O .

- _____ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- _____ 12. An acid contains one or more hydroxide ions.
- _____ 13. Chemists have a special system for naming bases.

1 Describing Acids And Bases Answers

John Stranger Holman, Philip Stone



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