Chapter 1 - Chemistry: The Study of Change

Element, compound, homogeneous mixture (solution), or heterogeneous mixture:

a) orange juice dreshly-squeezed)

b) brass

c) 0.9% saline (NaCl) solution

d) garden soil

e) room air

f) methane gas

g) sodium metal

h) Nagas

i) Cu(NO₃)₂ crystals

Define (some of these terms are found in Chapters 2 and 3):

a) element

c) solution

e) chemical change

g) intensive properties

i) precision.

k) isotopes

m) the mole

o) limiting reactant

b) compound

d) physical change

f) law of conservation of mass

h) extensive properties

j) accuracy

stoichiometry

n) Avogadro's number

Convert

a) 5 feet 10 inches to meters

b) 55 miles per hour to cm per second

c) 5.0 quarts to liters

d) 1.00 ft² to cm²

e) 32.06 amu to g

f) 25 g of carbon to moles of carbon

g) 1.00 g of gold to number of atoms of gold

h) 25 F to Kelvins

4. How many significant figures are in the following numbers?

a) 3.1416

b) 6.022 x 10²³ atoms/mole

c) 0.000104 g

d) 6.0 g/mL

e) 1.00794 amu

f) 5000 mi

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John Obimakinde,Samuel
Obimakinde,Ebenezer
Obimakinde,Fredrick Akinbolade

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