

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the processes of a reaction.
- 2. Explain how to write a balanced equation.
- 3. List the steps for writing a complete ionic equation.

Vocabulary

- 1. chemical equation
- 2. balanced equation
- 3. catalyst
- 4. coefficient
- 5. skeletal equation

Part A Completion

Write the coefficient for each in the balanced equation, and balance the equation by changing the coefficient for each substance. Write balanced below the equation. (Write 1 in the blank.)

_____ of sodium carbonate is reacting with _____ of calcium hydroxide. The products are _____ of sodium hydroxide and _____ of calcium carbonate. The new compound is _____, a chemical reaction between two substances. _____ is placed _____ in the equation to indicate that the high number of atoms of the _____ ions is found in the equation. The equation was written before the _____ of the chemical reaction was known.

The chemical reaction between _____ and _____ is a reaction. The equation is _____ that a solid _____ and a gas _____ are produced. _____ is placed _____ of _____ to show that the reaction is a chemical reaction. The equation is _____.

1. _____
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