Directions: Match each item with the correct statement below.

a. +

e. (aq) f. \xrightarrow{h} g. \xrightarrow{h} h. (g)

1. _____ Indicates a use of a catalyst in a reaction.

2. ____ Designates a reactant or product in the liquid state

b. →

Used to separate two reactants or two products
 Designates a reactant or product in the solid state

Indicates that heat is supplied to the reaction
 Designates a reactant or product in the gaseous state

Indicates that a reactant or product is dissolved in water

8. _____ "Yields," separates reactants from products

Directions: For each question, read the statement and select the best response. Write the letter of your answer on the line preceding the question.

 A ___ is a representation of a chemical reaction; the formula(s) of the reactant(s) are connected by an arrow to the formula(s) of the product(s).

a. catalyst

b. coefficient

c. chemical equation

d. mole

10. _____ The statement, "methane reacts with molecular oxygen to produce carbon dioxide and water" is an example of a ____.

d. (I)

a. skeletal equation

b. balanced equation

c. word equation

d. coefficient

In the balanced chemical equation below, the product(s) is/are ____. Select all that apply!
 BaCO₁₀₀ ___*+CO₂₀₀ + BaO₀₀

a. barium carbonate

b. carbon dioxide

e. barium oxide

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Prentice Hall Science Explorer: Chemical Interactions Pearson/Prentice Hall, Michael J. Padilla, 2004-10 Pearson Edexcel A Level Chemistry (Year 1 and Year 2) Andrew Hunt, Graham Curtis, Graham Hill, 2019-07-15 Develop and assess your students knowledge and skills throughout A level with worked examples practical assessment guidance and differentiated end of topic questions in this updated all in one textbook for Years 1 and 2 Combining everything your students need to know for the Pearson Edexcel A level Chemistry specification this revised textbook will Identify the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Student Book Provide support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Improve mathematical skills with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offer plenty of practice with Test yourself questions to help students assess their understanding and measure progress Encourage further reading and study with short passages of extension material Develop understanding with free online access to Test yourself answers Descriptive Inorganic Chemistry James E. House, Kathleen A. House, 2010-09-22 Descriptive and an extended glossary Inorganic Chemistry Second Edition covers the synthesis reactions and properties of elements and inorganic compounds for courses in descriptive inorganic chemistry This updated version includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes and incorporates new industrial applications matched to key topics in the text It is suitable for the one semester ACS recommended course or as a supplement in general chemistry courses Ideal for majors and non majors the book incorporates rich graphs and diagrams to enhance the content and maximize learning Includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes Incorporates new industrial applications matched to key topics in the text Focus on Physical Science California Edition Michael J. Padilla,2007 **Elements of Chemical Reaction Engineering** H. Scott Fogler, 2020-08-18 The Definitive Guide to Chemical Reaction Engineering Problem Solving With Updated Content and More Active Learning For decades H Scott Fogler's Elements of Chemical Reaction Engineering has been the world's dominant chemical reaction engineering text This Sixth Edition and integrated Web site deliver a more compelling active learning experience than ever before Using sliders and interactive examples in Wolfram Python POLYMATH and MATLAB students can explore reactions and reactors by running realistic simulation experiments Writing for today s students Fogler provides instant access to information avoids extraneous details and presents novel problems linking theory to practice Faculty can flexibly define their courses drawing on updated chapters problems and extensive Professional Reference Shelf web content at diverse levels of difficulty The book thoroughly prepares undergraduates to apply chemical reaction kinetics and physics to the design of chemical reactors And four advanced chapters address graduate level topics including effectiveness factors To support the field s growing emphasis

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