

14

THE BEHAVIOR OF GASES

Chapter Quiz

Indicate whether the statement is true or false.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to its pressure.
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to its temperature in kelvins.
3. According to Avogadro's law, the volume of a gas is directly proportional to the number of moles of gas.
4. Two different gases at the same volume, pressure, and temperature contain the same number of molecules.
5. The density of a gas is more closely related to its molecular weight than to its molecular volume.

Circle the letter of the correct answer for each question.

6. A gas expands from 1.0 liter to 2.0 liters at a constant pressure. What is the final temperature?

Answers:

1. True. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to its pressure.

Answers:

6. A. 2.0 K. According to Charles's law, the volume of a fixed amount of gas is directly proportional to its temperature in kelvins. If the volume doubles, the temperature must also double.

1. _____	True
2. _____	True
3. _____	True
4. _____	True
5. _____	True

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Unveiling the Magic of Words: A Review of "**1the Behavior Of Gases Chapter Quiz**"

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