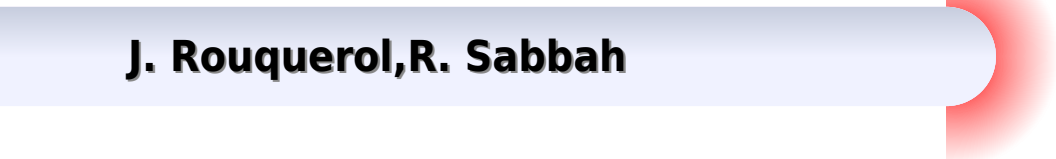


Thermochemical Equations

- **Thermochemical equations** are balanced chemical equations that include the physical states of all reactants and products and the energy change.
- $2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \quad \Delta\text{H} = 572 \text{ kJ}$
- $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$
 $\Delta\text{H} = -802 \text{ kJ}$

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J. Rouquerol, R. Sabbah



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