

STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of H_2SO_4 according to the following equation:



2. Calculate the mass of NH_3 that can be produced from the reaction of 125 g of NCl_3 according to the following equation:



3. Identify the limiting reactant and determine the mass of CO_2 that can be produced from the reaction of 25 g of C_3H_8 with 75 g of O_2 according to the following equation:



4. How many grams of SO_2 are produced when 152 g of CS_2 reacts with 48.0 g of O_2 according to the following equation:



Answers For The Chemistry Lab Of Stoichiometry

James F. Hall



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oiling 370ml water 75g pitted kalamata black olives 1 small red onion 10 cherry tomatoes 1 tbsp dried oregano about 150ml

olive oil to drizzle flaky sea salt method 1

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put the loaf tray on the middle shelf and bake for 25 minutes then lower the oven to 200 c 392 f gas mark 6 and bake for 10

15 minutes more until the crust has a good colour hold the loaf in a tea towel and tap the bottom if it sounds hollow it s

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method mix all the ingredients in a large bowl taking care not to put

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meanwhile heat the oven to 230 c 210 c fan 450 f gas 8 and put a roasting tray in the bottom of the oven to heat up step 9 remove the risen loaf from the bag and dust with flour fill the roasting tray with cold water to create steam and put the bread in the oven bake for 15 minutes then lower the oven to 190 c 170 c fan 375 f gas 5

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