

Name \_\_\_\_\_

Date \_\_\_\_\_

### Reading the Periodic Table **KEY**

1. List the atomic numbers of the elements in Period 2.

**3, 4, 5, 6, 7, 8, 9, 10**

2. List the symbols of the elements in Period 3.

**Na, Mg, Al, Si, P, S, Cl, Ar**

3. Name the elements in Period 2 that are metals.

**Li, Be**

4. Name the elements in Period 1 that are nonmetals.

**H, He**

5. Which element in Period 3 has the least mass?

**Sodium (Na)**

6. Which element in Period 4 has the most mass?

**Krypton (Kr)**

7. To which group/family do these belong?

a. Cobalt **transition metals**

b. Sodium **alkali metals**

c. Argon **noble gases**

d. Silicon **group 14 metalloids**

e. Chlorine **halogens**

f. Magnesium **alkaline earth metals**

8. Find iron on the periodic table:

a. What is the atomic number of iron? **26**

b. To which group does iron belong? **Transition metals**

9. Name the elements that have properties similar to that of chlorine.

**Fluorine, bromine, iodine, astatine, tennessine**

10. Fill in the chart below. The first row has been completed as an example.

| Element           | Symbol       | Atomic Number | Period       | Group         |
|-------------------|--------------|---------------|--------------|---------------|
| sodium            | Na           | 11            | 3            | 1             |
| <del>oxygen</del> | <del>O</del> | <del>8</del>  | <del>2</del> | <del>16</del> |
| krypton           | Kr           | 36            | 4            | 18            |
| gold              | Au           | 79            | 6            | 11            |
| cadmium           | Cd           | 48            | 5            | 12            |
| manganese         | Mn           | 25            | 4            | 7             |

# Answer Key To The Periodic Table Homework

**George Graybill**



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