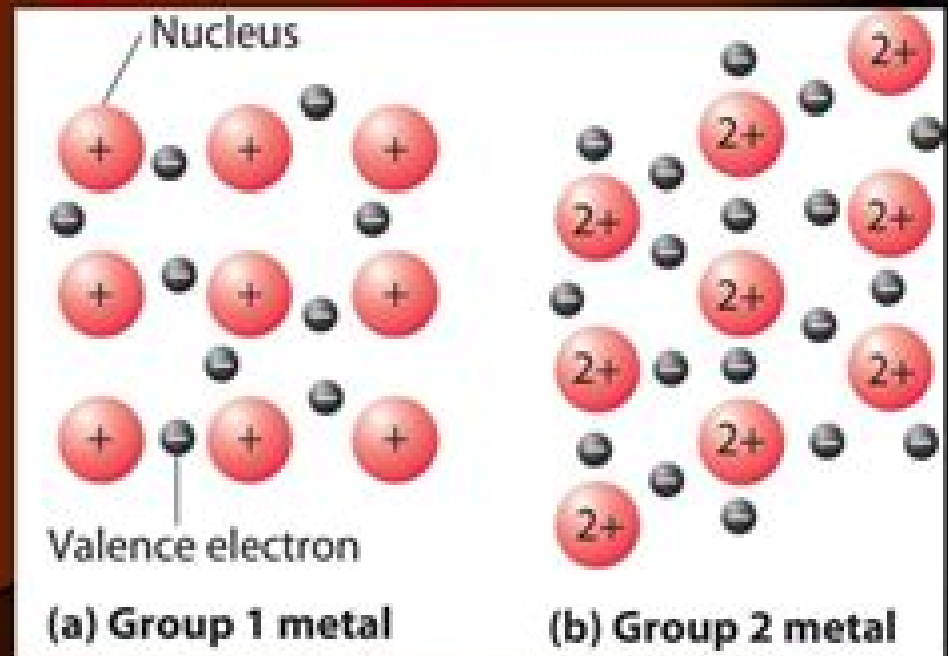


# Metallic bonding

- What happens when a metal bonds with another metal?
- Both metal atoms give up their valence electrons.
- This forms a "sea of electrons" that fills the space between the metal cations. The attraction of the positively charged metal cations to the negative valence electrons is what holds the metal together.



# Answers For Ionic Metallic Bonding

**B.K. Sharma**



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