or indirectly before the examination. In addition to the fittings and reagents normally contained in a Chemistry laboratory, the following apparatus and materials will be 2. required by each candidate:

(a) one burette of 50 cm3 capacity;

(b) one pipette, either 20 cm3 or 25 cm3. All candidates at one centre must use pipettes of the same volume. These should be clean and free from grease;

(c) the usual apparatus for titration;

(d) the usual apparatus and reagents for qualitative work including the following with all reagents appropriately labelled:

dilute sodium hydroxide solution; (i)

dilute hydrochloric acid; (iii)

dilute trioxonitrate (V) acid; (iiii)

silver trioxonitrate (V) solution; (iv)

aqueous barium chloride: (v)

aqueous ammonia: (vi)

lime water: (viii)

(viii) red and blue litmus papers;

dilute tetraoxosulphate (VI) acid; (ix)

acidified potassium dichromate solution. (x)

(e) spatula:

(f) filtration apparatus;

(g) one beaker:

(h) one boiling tube;

(i) four test tubes:

(i) methyl orange indicator;

(k) glass rod:

(1) wash bottle containing distilled/deionized water;

(m) burning splint;

(n) watch glass;

(o) bunsen burner/source of heat;

(p) droppers

(q) mathematical table/calculator;

(r) phenolphthalein indicator.

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Eric John Holmyard

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Annual Conference Proceedings Science Teachers Association of Nigeria. Conference, 1999 West African Journal of Education, 1970 The Farmer & Stock-breeder, 1948 British Book News, 1988 Cumulative Author Index to Psychological Abstracts, 1978 Advanced Practical Chemistry, 1985 Practical Chemistry, Part I Edmund Arthur O'Connor, 1949 Practical Chemistry Eric John Holmyard, 1923

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