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web 1 place a glass of water outside it will freeze if the temperature is below 0 c 3 a law states a consistently observed phenomenon can be used for prediction b theory a widely accepted explanation of the behavior of matter c hypothesis a tentative explanation can be investigated by experimentation 5

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web hclo 4 is the stronger acid because in a series of oxyacids with similar formulas the higher the electronegativity of the central atom the stronger is the attraction of the central atom for the electrons of the oxygen s the stronger attraction of the oxygen electron results in a stronger attraction of oxygen for the electrons in the o h bond making the hydrogen

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web an ion that is found in both salts in a solution common ion effect the lowering of the solubility of an ionic compound as a result of the addition of a common ion is called this lower the smaller the numerical value of the solubility product constant the the solubility of the compound precipitate

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contain a maximum of two electrons differences σ bonds are stronger and result from end to end overlap and all single bonds are σ bonds π bonds between the same two atoms are weaker because they result from side by side

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web here the value of concentration of water is taken as 0 36 instead of 0 16 divide the product of the concentration of water carbon monoxide and the equilibrium constant by the concentration of hydrogen instead of multiplying it correct d the value of K_{eq} is taken as 4 4 instead of 0 44