

13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
  - Two nonmetal atoms bonded together
  - Two metal atoms bonded together
  - A metal and a nonmetal atom bonded together
  - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
  - Covalent
  - Both a and b
  - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
  - Covalent
  - Both a and b
  - Neither a nor b
16. (4 pts) How many total atoms are in this molecule:  $C_6H_{12}OH$ ?
- 3
  - 4
  - 19
  - 20
  - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is  $C_5H_{10}Cl_x$ , what must  $x$  be?
- 1
  - 2
  - 3
  - 4
  - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water?  $C_3H_6ClF$   $C_6H_5CH_3$   $CS_2$   $BaSO_4$
- $C_3H_6ClF$
  - $C_6H_5CH_3$
  - $CS_2$
  - $BaSO_4$
  - $CS_2$  and  $BaSO_4$

# Answers For Apologia Chemistry Module 15 Test

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## **Table of Contents Answers For Apologia Chemistry Module 15 Test**

1. Understanding the eBook Answers For Apologia Chemistry Module 15 Test
  - The Rise of Digital Reading Answers For Apologia Chemistry Module 15 Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Apologia Chemistry Module 15 Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answers For Apologia Chemistry Module 15 Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Apologia Chemistry Module 15 Test
  - Personalized Recommendations
  - Answers For Apologia Chemistry Module 15 Test User Reviews and Ratings

- Answers For Apologia Chemistry Module 15 Test and Bestseller Lists
- 5. Accessing Answers For Apologia Chemistry Module 15 Test Free and Paid eBooks
  - Answers For Apologia Chemistry Module 15 Test Public Domain eBooks
  - Answers For Apologia Chemistry Module 15 Test eBook Subscription Services
  - Answers For Apologia Chemistry Module 15 Test Budget-Friendly Options
- 6. Navigating Answers For Apologia Chemistry Module 15 Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Answers For Apologia Chemistry Module 15 Test Compatibility with Devices
  - Answers For Apologia Chemistry Module 15 Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answers For Apologia Chemistry Module 15 Test
  - Highlighting and Note-Taking Answers For Apologia Chemistry Module 15 Test
  - Interactive Elements Answers For Apologia Chemistry Module 15 Test
- 8. Staying Engaged with Answers For Apologia Chemistry Module 15 Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answers For Apologia Chemistry Module 15 Test
- 9. Balancing eBooks and Physical Books Answers For Apologia Chemistry Module 15 Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answers For Apologia Chemistry Module 15 Test
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Apologia Chemistry Module 15 Test
  - Setting Reading Goals Answers For Apologia Chemistry Module 15 Test
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Apologia Chemistry Module 15 Test
  - Fact-Checking eBook Content of Answers For Apologia Chemistry Module 15 Test
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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