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ppm Answer-  $(\text{grams analyte}/\text{grams sample}) \times 10^6$

Molarity Answer- moles analyte/liter of solution

Volume Percent Answer-  $(\text{volume solute}/\text{volume solution}) \times 100$

Volume ppm Answer-  $(\text{volume solute}/\text{volume solution}) \times 10^6$

kilo- Answer-  $10^3$

deci- Answer-  $10^{-1}$

centi- Answer-  $10^{-2}$

milli- Answer-  $10^{-3}$

micro- Answer-  $10^{-6}$

nano- Answer-  $10^{-9}$

pico- Answer-  $10^{-12}$

femto- Answer-  $10^{-15}$

weight percent Answer-  $(\text{grams analyte}/\text{grams sample}) \times 100$

ppt Answer-  $(\text{grams analyte}/\text{grams sample}) \times 10^3$

ppt simplified Answer- gram analyte/liter solution

# Analytical Chemistry Exam Study Guide

**Arshad Iqbal**



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