# Naming Acids and Bases Answers

Name the following acids and bases:

1)	NaOH	sodium	hydroxide

- 2) H<sub>2</sub>SO<sub>3</sub> sulfurous acid
- 3) H<sub>2</sub>S <u>hydrosulfuric acid</u>
- 4) H<sub>3</sub>PO<sub>4</sub> phosphoric acid
- 5) NH<sub>3</sub> ammonia
- 6) HCN hydrocyanic acid
- 7) Ca(OH)<sub>2</sub> calcium hydroxide
- 8) Fe(OH)<sub>3</sub> iron (III) hydroxide
- 9) H<sub>3</sub>P <u>hydrophosphoric acid</u>

Write the formulas of the following acids and bases:

- 10) hydrofluoric acid HF
- hydroselenic acid H<sub>2</sub>Se
- carbonic acid H<sub>2</sub>CO<sub>3</sub>
- 13) lithium hydroxide LiOH
- 14) nitrous acid HNO<sub>2</sub>
- cobalt (II) hydroxide <u>Co(OH)</u>
- sulfuric acid H₂SO₄
- beryllium hydroxide <u>Be(OH)</u>
- 18) hydrobromic acid HBr

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**Darshan Singh Sarai** 

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