Name KEY Pr. Date Write balanced formula equations for these reactions and identify the type of reaction. ▶2KI Potassium + Iodine Synthesis \_\_ 2K + I2 Barium Iodide + Sodium Sulfate \_\_\_\_\_\_ double rep BaIz + Na 2 SO4 -> Ba SO4 + 2NaI Aluminum Hydroxide decomposition 2 AI(OH)3 -> AI2O3 + 3H2O 4. Sodium Chlorate NaCl + Oz decomposition Aluminum Bromide + Chlorine ---single replacement AIBry + Cl2 -> AICI3+ Bry 6. Phosphoric Acid + Magnesium ----H3PO4 + Mg -> Mq3(PO4)2+ H2 7. Iron III Sulfate + Calcium Nitrate ---double rep. Fez(Say)3 +3Ca(NO3)2 -> 2Fe(NO3)2+3Ca SO4 8. Sodium + Water w Stragte vep ZNa+2H2O -> ZNaOH + HZ 9. Aluminum + Iodine ---synthesis 2 AL +3T2 -> ZALT2 10. Iron III Hydroxide + Hydrobromic Acid double vep. Fe (OH), +3HBr -> FeBr, +3HOH Magnesium Hydroxide + Ammonium Sulfate — double rep Mg(OH). + (NH4) SO4 -> MgSG4 + 2NH4OH 12. Water -12H2+ Dx decomposition 2420 13. Calcium + Hydrochloric Acid ---single repassment ca +2HCI -> Cacis + Hz 14. Lithium + Chlorine -Synthesis 2 Li+Ch -> 2 LiC1

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**Martha Mackin** 

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Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students GED Test Prep Plus 2020 Caren Van Slyke, 2019-12-03 Always study with the most up to date prep Look for GED Test Prep Plus 2021 ISBN 9781506266251 on sale December 01 2020 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entitles included with the product Clinical Chemistry: Principles, **Techniques, and Correlations, Enhanced Edition** Michael L. Bishop, 2020-06-11 Clinical Chemistry Principles Techniques and Correlations Enhanced Eighth Edition demonstrates the how what why and when of clinical testing and testing correlations to help you develop the interpretive and analytic skills you ll need in your future career

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