Chemistry 105 Practice Exam 1

Multiple Choice Section

I.An el	lement		
m.	can be separated into its components by p	bystant.	methods.
ča.	always has the same chemical properties regardless of its source.		
65	cannot be separated into simpler substances by chemical methods.		
di.			
di-	exists only as atoms, not as molecules.		
2.W%6	ch one of the following represents a chemica	t change	e?
OL.	Mercury (II) oxide is heated to and forms mercury metal and oxygen gas.		
ās.			
65			
di.			
di-	Rock salt is pulverized.		
S. Equi	al amounts of water are present under the fo	diowing.	conditions. In which case do the water molecules have th
h-6	ghest kinetic energy?		
a.	as fee at -10°C (14°F)	- d.	in the gas phase at 150°C (302°F)
250	as steam at 100°C (212°F)		in the solid phase at 0°C (32°F)
c.	In the liquid phase at 80°C (176°F)		
4.19766	ch of the following represents the largest ma	us 7	
80	250 ng		0.25 kg
	25 µg	45	25 mg
C.	2.5 g		
716		and 2.5	sity of a sample of seawater and obtains the following 121 g/mL. Known values are reported to be in the 1.02 to —
a.	less precise than the known values.	· A.	precise and accurate.
Sa.	more accurate than the known values.	65	Inaccurate but precise.
6	Imprecise but accurate.		
6.Acco	rding to the law of definite proportions,		
G.	atoms forming a given compound react in	variabl	e proportions depending on conditions.
ās.			he same proportions of the same elements
6	all compounds containing the same types		
d.	all compounds containing the same types multiples.	of atoms	s have relative masses that are whole-number

7. Which of the following statements is NOT true regarding blackbody radiation, the ultraviolet catastrophe, and Planck's view of electromagnetic radiation?

e. only one type of molecule can be produced when two elements combine.

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Dr. B. Kapila, S. K. Khanna

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Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design **Ebook: Introductory Chemistry: An Atoms First Approach** Burdge, 2016-04-16 Ebook Introductory Chemistry An Atoms First Approach Calculating Theoretical Yield Is Not Impossible! Michael S. Leonard, 2025-02-25 Calculating theoretical and percent yield is a fundamental skill for the laboratory This book primarily targets Organic Chemistry Laboratory courses at the high school or college and university level as a supplemental resource to help students master this skill It begins with simple examples from everyday life demonstrates the importance of balancing the equation addresses the role of the mole in these computations discusses different types of liquids considers the role of significant figures and culminates with the planning of syntheses There are suggestions for further reading as well as practice problems and questions to ensure mastery Begins with examples from everyday life that enable students to understand the concepts of theoretical and percent yield before applying those concepts to the laboratory Addresses the necessity of balancing the reaction equation the centrality of the mole in these calculations and the role of significant figures in reporting the answer Explains how to approach the calculations when using neat liquids or solutions The culmination of this text is the use of the same thought processes to plan the amounts of reactants needed for syntheses of desired quantities of product All of the problems in the book include detailed solutions with accompanying text to explain the answers and ancillaries also include Chapter-wise DPP Sheets for Chemistry NEET Disha Experts, 2019-10-10 The book suggestions for further reading Chapter wise Daily Practice Problem DPP Sheets for Chemistry NEET contains 1 Carefully selected Questions 45 per DPP in Chapter wise DPP Sheets for Practice 2 The book is divided into 30 Chapter wise DPPs based on the NCERT 3 Time Limit Maximum Marks Cutoff Qualifying Score for each DPP Sheet is provided 4 These sheets will act as an Ultimate tool for Concept Checking Speed Building 5 Collection of 1395 MCQ s of all variety of new pattern 6 Covers all important Concepts of each Chapter 7 As per latest pattern syllabus of JEE Main exam Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-01-19 The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms. The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the

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