

WATER AND AQUEOUS SYSTEMS

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LIQUID WATER AND ITS PROPERTIES (pages 475-478) SECTION 17.1

This section explains the hydrogen bonding that occurs in water molecules and how high surface tension and low vapor pressure can be explained in terms of hydrogen bonding.

➤ The Water Holecule (pages 475-476)

- What unique molecular compound is the foundation of all life on Earth? water.
- 2. Approximately what fraction of Earth's surface is covered in water?
- 3. Circle the letter next to each sentence that is true concerning water molecules.
 - Each O H covalent bond in a water molecule is nonpolar.
 - b. In a water molecule, the less electronegative hydrogen atoms acquire a slight positive charge and the oxygen atom acquires a slight negative change.
 - c. Because the water molecule has an H O H bond angle of 105°, the molecule as a whole is polar.
- 4. The diagram below depicts a water molecule. Complete the labels showing the locations of the hydrogen atoms, the oxygen atom, and the regions of positive and negative charge.



List five important properties of water that result from hydrogen bonding.

high surface tension.

low vapor pressure

high specific heat capacity

high heat of vaporization

high boiling point

1water And Aqueous Systems Guided Answers

Thomas Griffiths

1water And Aqueous Systems Guided Answers:

Unveiling the Magic of Words: A Report on "Iwater And Aqueous Systems Guided Answers"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really aweinspiring. Enter the realm of "**1water And Aqueous Systems Guided Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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