# Overview of Chemical Bonding and Chemical Reactions

Fill in the blanks with the correct information about chemical bonding.

An \_\_\_\_ is an atom that has either lost or gained an electron.

An atom becomes \_\_\_\_\_ only when its outermost shell is filled with electrons.

A chemical bond is formed when atoms or electrons.

 An \_\_\_\_\_\_ is a force of attraction between opposing charges of the ions in the compound.

 A \_\_\_\_\_\_ involves two or more electrons sharing electrons to achieve stability.

6. A molecule has a positive and a negative end.

7. A non-polar molecule is one whose constituent atoms electrons equally.

8. A ion is a group of atoms with a positive or negative charge.

NaCl is an compound.

is the property that determines whether or not a molecule is polar.

do not easily form bonds, as their outermost shells are filled.

12. The number of electrons in the outermost shell of an atom \_\_\_\_\_ as we go left to right in a period.

13. The reactivity of halogens \_\_\_\_\_ as we move down their group.

14. H<sub>2</sub> is a \_\_\_\_\_ covalent compound.

15. bonds are formed when metals share their electrons.

16. A \_\_\_\_\_ bond has two pairs of electrons.

# 2 3 Chemical Reactions And Bonding Worksheet Answers

**Chao Zhang** 

2 3 Chemical Reactions And Bonding Worksheet Answers:			

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