

## 7.2 PhET LAB: Alpha Decay



Name: \_\_\_\_\_

**OBJECTIVE:** explain the alpha decay process; complete radioactive decay equations; define and analyze half-life through applying the PhET "Alpha Decay" simulation at <http://phet.colorado.edu>.

Open/Run Alpha Decay at <http://phet.colorado.edu>.

Take 5 minutes to freely explore the sim.

### Investigating Alpha Decay

1. Start on the **SINGLE ATOM** tab. Observe the decay of Polonium-211. After each decay, press the RESET NUCLEUS button to watch the process again. Write a description of alpha decay for Po-211.

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Complete the following alpha decay equations, using <http://www.chemicalelements.com> as a resource:



7. How is alpha decay used in everyday life? (give at least two uses)

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### Investigating Half-Life of Alpha Decay

8. Click the **MULTIPLE ATOMS** tab. Execute five trials to determine the number of parent and daughter nuclei at one half-life. Complete the table below.

| Parent Nuclei<br>Po-211 | Parent Nuclei<br>(remaining nuclei)<br>[yellow] | Daughter Nuclei<br>(decayed nuclei)<br>[black] |
|-------------------------|-------------------------------------------------|------------------------------------------------|
| 100                     |                                                 |                                                |
| 80                      |                                                 |                                                |
| 60                      |                                                 |                                                |
| 40                      |                                                 |                                                |
| 20                      |                                                 |                                                |

9. Define half-life.

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# Alpha Decay Phet Answers

**William Patton**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

## Alpha Decay Phet Answers:

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**Alpha Decay in Isotopes of Atomic Number Less Than 83** ,1949 Some time ago we started work in an attempt to observe alpha particle decay in isotopes of atomic number less than 83 In the first experiments thin targets of gold leaf were bombarded with 190 Mev deuterons in the 184 inch cyclotron Two alpha decay periods were observed in these targets one of 0.7 minutes half life and another of 4.3 minutes half life The alpha particle energies were 5.7 and 5.2 Mev respectively Chemical separations proved that the 4.3 minute period is due to a gold isotope and suggested that the 0.7 minute period is due to a mercury isotope The mass numbers of these new isotopes have not been determined However the results of excitation functions in the production of the gold isotope by bombarding gold and platinum with protons suggest that its mass number lies in the range 185-188 The work on this isotope indicates that the alpha to electron capture branching ratio is of the order of magnitude of 10<sup>-4</sup> and that positron activity accompanies the 4.3 minute alpha period      **Alpha Decay as a Fission-like Process** D. N. Poenaru,M. Ivascu,A. Sandulescu,1979      **Alpha Decay as a Weak Interaction** A. O. Barut,A. Sevgen,1973      *On the First Structure in Alpha Decay* Age Bohr,P.O. Fröman,Ben R. Mottelson,1955      **Alpha Decay from Deformed Nuclei** Tore Berggren,Peter Olanders,1987      *Systematics of Alpha Decay Half-life* ,2006      Alpha Decay Joseph J. Devaney,1950      Alpha Decay and Excited States of <sup>111</sup>Xe Luigi Capponi,2014      **Theory of Alpha Decay of Spheroidal Nuclei** John O. Rasmussen Jr,1953

## Unveiling the Power of Verbal Art: An Emotional Sojourn through **Alpha Decay Phet Answers**

In a world inundated with monitors and the cacophony of fast interaction, the profound power and mental resonance of verbal beauty usually disappear into obscurity, eclipsed by the regular assault of sound and distractions. However, nestled within the lyrical pages of **Alpha Decay Phet Answers**, a interesting function of fictional elegance that impulses with natural feelings, lies an memorable journey waiting to be embarked upon. Written with a virtuoso wordsmith, this interesting opus courses viewers on an emotional odyssey, lightly exposing the latent possible and profound affect embedded within the complicated web of language. Within the heart-wrenching expanse of this evocative evaluation, we shall embark upon an introspective exploration of the book is key subjects, dissect its fascinating publishing type, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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