



PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2

TIME: 1 HOUR

VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Table of Contents 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper

1. Understanding the eBook 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - The Rise of Digital Reading 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper

- Personalized Recommendations
 - 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper User Reviews and Ratings
 - 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper and Bestseller Lists
5. Accessing 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper Free and Paid eBooks
- 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper Public Domain eBooks
 - 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper eBook Subscription Services
 - 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper Budget-Friendly Options
6. Navigating 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper eBook Formats
- ePub, PDF, MOBI, and More
 - 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper Compatibility with Devices
 - 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - Highlighting and Note-Taking 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - Interactive Elements 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
8. Staying Engaged with 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
9. Balancing eBooks and Physical Books 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
- Setting Reading Goals 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper

- Fact-Checking eBook Content of 14 Physical Science Term 1 Controlled Test Grade 12 Commom Paper
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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