Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

Service	zinc	iron (II)	iron (III)	gallium	silver	lead (IV)
chloride	ZnCl ₂	FeCl ₂	FeCl ₃	GaCl ₃	AgCI	PbCI ₄
acetate	Zn(C ₂ H ₁ O ₂) ₂	Fe(C ₂ H ₂ O ₂) ₂	Fe(C ₂ H ₃ O ₂) ₃	Ga(C ₂ H ₃ O ₂) ₃	Ag C ₂ H ₃ O ₂	Pb(C ₂ H ₁ O ₂) ₄
nitrate	Zn(NO ₃) ₂	Fe(NO ₃) ₂	Fe(NO ₃) ₃	Ga(NO ₃) ₃	AgNO ₃	Pb(NO ₃) ₄
oxide	ZnO	FeO	Fe ₂ O ₃	Ga ₂ O ₃	Ag ₂ O	PbO ₂
nitride	Zn ₃ N ₂	Fe ₃ N ₂	FeN	GaN	Ag ₂ N	Pb ₃ N ₄
sulfate	ZnSO ₄	FeSO ₄	Fe ₂ (SO ₄) ₃	Ga ₂ (SO ₄) ₃	Ag ₂ SO ₄	Pb(SO ₄) ₂

Write the formulas for the following compounds:

- copper (II) chloride CuCl₂
- lithium acetate LiC₂H₃O₂
- vanadium (III) selenide V₂Se₃
- 4) manganese (IV) nitride Mn₃N₄
- 5) beryllium oxide BeO
- sodium sulfate Na₂SO₄
- 7) aluminum arsenide AlAs
- potassium permanganate KMnO₄
- 9) chromium (VI) cyanide Cr(CN)₆
- 10) tin (II) sulfite SnSO₃
- 11) vanadium (V) fluoride VF₅
- ammonium nitrate NH₄NO₃

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JL Elias

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Table of Contents Answers For Ionic Formula Writing

- 1. Understanding the eBook Answers For Ionic Formula Writing
 - The Rise of Digital Reading Answers For Ionic Formula Writing
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Answers For Ionic Formula Writing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Ionic Formula Writing
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answers For Ionic Formula Writing
 - Personalized Recommendations
 - Answers For Ionic Formula Writing User Reviews and Ratings
 - Answers For Ionic Formula Writing and Bestseller Lists
- 5. Accessing Answers For Ionic Formula Writing Free and Paid eBooks

- Answers For Ionic Formula Writing Public Domain eBooks
- Answers For Ionic Formula Writing eBook Subscription Services
- Answers For Ionic Formula Writing Budget-Friendly Options
- 6. Navigating Answers For Ionic Formula Writing eBook Formats
 - o ePub, PDF, MOBI, and More
 - Answers For Ionic Formula Writing Compatibility with Devices
 - Answers For Ionic Formula Writing Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Ionic Formula Writing
 - Highlighting and Note-Taking Answers For Ionic Formula Writing
 - Interactive Elements Answers For Ionic Formula Writing
- 8. Staying Engaged with Answers For Ionic Formula Writing
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Ionic Formula Writing
- 9. Balancing eBooks and Physical Books Answers For Ionic Formula Writing
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Ionic Formula Writing
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Ionic Formula Writing
 - Setting Reading Goals Answers For Ionic Formula Writing
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Ionic Formula Writing
 - Fact-Checking eBook Content of Answers For Ionic Formula Writing
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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