

## Vocabulary

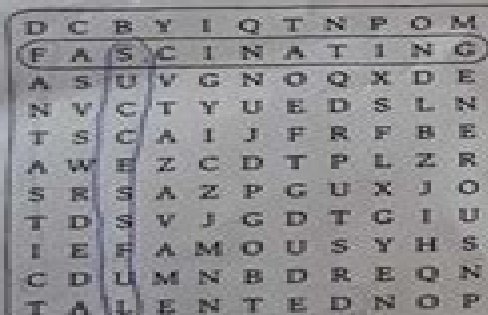
- 1 Complete the sentences. Use the words in the box.

documentary film programmes quiz  
satellite advertisement

- 0 I enjoy watching programmes about different countries.  
1 There's a good ..... on Saturday evening - people answer questions and win money.  
2 I saw a new ..... for a mobile phone yesterday. It was great.  
3 My favourite ..... is about life in the eighteenth century.  
4 The ..... Mission Impossible is on tonight, but it starts very late.  
5 How many channels can you receive on your ..... TV?

☐ 5

- 2 Find five more adjectives.



- 0 fascinating.  
1 .....  
2 .....  
3 .....  
4 .....  
5 .....

☐ 5

## Language in Context

- 3 Match 1-5 with A-E.

- 0 Can you keep a promise? .....  
1 It's raining. Shall we watch TV? .....  
2 Are we going to the cinema tonight? .....  
3 Is Bill Gates American? .....  
4 We can't go on the picnic. It's raining. ....  
5 I want a chicken and jam sandwich. ....

- A I hope so.  
B What a silly idea!  
C That's right.  
D Oh dear!  
E I don't know. What's on?  
F Yes, of course I can.

☐ 5

- 4 Circle the correct answer - A, B or C.

- 0 Where's Jack?  
A He's having breakfast.  
B He has breakfast.  
C He having breakfast.  
1 What did you think of the documentary about Wales?  
A It was talented.  
B It was generous.  
C It was fascinating.  
2 Let's watch this quiz! It's fantastic.  
A Oh, no! I hate it.  
B Oh dear! I love it.  
C That's brilliant! I hate it.  
3 The English test was difficult. I think I did really badly.  
A I hope so.  
B Oh dear!  
C What a silly idea!  
4 Look - it's my new camera!  
A It's great. Can I borrow it?  
B It's great. Can I have it?  
C It's great. Can I take it?  
5 Did you see the film last night?  
A No, I'm studying.  
B No, I was studying.  
C No, I studied.

☐ 5

# Answers For Pearson Education

**Sue Hart**



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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with hcl as 57.9 kJ mol<sup>-1</sup>

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic

eg 1  $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$  for which  $\Delta H = 57 \text{ kJ mol}^{-1}$  eg 2  $\text{HNO}_3(\text{aq}) + \text{KOH(aq)} \rightarrow \text{KNO}_3(\text{aq}) + \text{H}_2\text{O(l)}$  for which  $\Delta H = 57 \text{ kJ mol}^{-1}$

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example  $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(aq)}$  you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts

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