

CHEMICAL REACTIONS AND EQUATIONS

Objective Questions

Multiple Choice Questions

1. Which of the statements about the reaction below are incorrect ?

$$2\text{PbO}(s) + \text{C}(s) \longrightarrow 2\text{Pb}(s) + \text{CO}_2(g)$$

- (i) Lead is getting reduced
- (ii) Carbon dioxide is getting oxidised
- (iii) Carbon is getting oxidised
- (iv) Lead oxide is getting reduced
- (a) (i) and (ii)
- (b) (i) and (iii)
- (c) (i), (ii) and (iii)
- (d) all of these.

2. $\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$

The above reaction is an example of a—

(2024)

- (a) combination reaction
- (b) double displacement reaction
- (c) decomposition reaction
- (d) displacement reaction.

3. What happens when dilute hydrochloric acid is added to iron fillings ? (2022)

- (a) Hydrogen gas and iron chloride are produced
- (b) Chlorine gas and iron hydroxide are produced
- (c) No reaction takes place
- (d) Iron salt and water are produced.

In the reaction $2\text{FeCl}_2 + \text{Cl}_2 \longrightarrow 2\text{FeCl}_3$ Chlorine is a—

- (a) Reducing agent
- (b) Oxidizing agent
- (c) Catalyst
- (d) Inert medium.

Rancidity can be prevented in food items by—

- (a) Mixing antioxidant
- (b) Keeping in refrigerator
- (c) Keeping away from light
- (d) All of these.

Which among the following is not an oxidizing agent ?

- (a) Oxygen
- (b) Conc. Sulphuric acid
- (c) Chlorine
- (d) Hydrogen.

Which among the following gases are used to keep the oily food items fresh ?

- (a) Carbon dioxide and Oxygen
- (b) Nitrogen and helium
- (c) Helium and Oxygen
- (d) Nitrogen and Oxygen.

Chemically rust is—

- (a) Hydrated ferrous oxide
- (b) Ferric oxide
- (c) Hydrated ferric oxide
- (d) None of these.

An exothermic reaction in which glucose combines with oxygen in the cells of our body and provides energy—

- (a) Respiration
- (b) Digestion
- (c) Excretion
- (d) None of these.

1chemical Reactions Answers

Kristin Kemp



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The Basics of Chemical Reactions Krista West,2013-12-15 The chemical reactions that shape the world are sometimes simple and sometimes complex which is at the very core of this informative text This volume explores the simplicity of basic chemical reactions and then builds to the more complex giving readers a history of the years and the minds that contributed to the research that led to chemistry as we know it today Biographical sidebars provide unique information about scientists who are valued in the field but are often not widely known

Chemistry in a Month John Wilkinson,2003

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