

Balancing Equations & Types of Reactions WS 1

Practice Problems

In your notebook, solve the following problems.

NOTE: A skeleton equation is an unbalanced equation.

SECTION 10.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.



2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.



3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.



4. Write a balanced equation for the production of HCl gas from its elements.



5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



Hydrochloric acid and calcium carbonate react to produce carbon dioxide, calcium chloride, and water.

6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



2 solid silver reacts w/ 1 solid sulfur to create solid silver sulfide.

1section Review Describing Chemical Reactions

BM King

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Patai's 1992 Guide to the Chemistry of Functional Groups Saul Patai, 1992-04-08 Patai's 1992 Guide to the Chemistry of Functional Groups Saul Patai The Hebrew University of Jerusalem Israel Ever since the publication of the first volume of The Chemistry of Functional Groups in 1964 the Patai series has acted as an essential reference source to many researchers By the end of 1991 the series consisted of 50 titles bound in 73 volumes containing nearly 900 chapters written by over 1250 authors The aim of this Guide as was that of the previous edition is to present sufficient material on each of the published chapters to allow the researcher to decide whether these chapters are relevant and useful for his or her purpose and thus worth pursuing in full For those who are familiar with only selected volumes from the series the cross referencing between complementary and related chapters from different volumes will be invaluable The Guide is fully indexed by both subject and author thus making it an essential reference tool for all organic chemists Bioconjugate Techniques Greg T. Hermanson, 2010-07-26 Bioconjugate Techniques 2nd Edition is the essential guide to the modification and cross linking of biomolecules for use in research diagnostics and therapeutics It provides highly detailed information on the chemistry reagent systems and practical applications for creating labeled or conjugate molecules It also describes dozens of reactions with details on hundreds of commercially available reagents and the use of these reagents for modifying or cross linking peptides and proteins sugars and polysaccharides nucleic acids and oligonucleotides lipids and synthetic polymers A one stop source for proven methods and protocols for synthesizing bioconjugates in the lab Step by step presentation makes the book an ideal source for researchers who are less familiar with the synthesis of bioconjugates More than 600 figures that visually describe the complex reactions associated with the synthesis of bioconjugates Includes entirely new chapters on the latest areas in the field of bioconjugation as follows Microparticles and nanoparticles Silane coupling agents Dendrimers and dendrons Chemoselective ligation Quantum dots Lanthanide chelates Cyanine dyes Discrete PEG compounds Buckyballs fullerenes and carbon nanotubes Mass tags and isotope tags Bioconjugation in the study of protein interactions **Prentice Hall Science Series, 1994** Prentice-Hall Staff, 1993 **Review of Public Comments on the October 1988 Draft of the "Plan and Schedule for NAPAP Assessment Reports, 1989-1990."** National Acid Precipitation Assessment Program (U.S.), 1989 CRC Critical Reviews in Biochemistry Chemical Rubber Company, 1972

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1. The book is structured into several chapters, namely:

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- Chapter 4: 1section Review Describing Chemical Reactions in Specific Contexts
- Chapter 5: Conclusion

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