

Chemistry Worksheets Class 10 on Chapter 2 Acids, Bases and Salts - Set 2

Q1. Fresh milk has a pH of 6. When milk changes into curd, the pH value will:

- a.) become 7
- b.) become less than 6
- c.) become more than 7
- d.) remains unchanged

Q2. Which of the following is medicine for indigestion?

- a.) sodium hydroxide
- b.) manganese hydroxide
- c.) magnesium hydroxide
- d.) potassium hydroxide

Q3. How many number of water of crystallisation is present in copper sulphate crystals?

- a.) 3
- b.) 5
- c.) 7
- d.) None

Q4. The indicators which turn red in acid solution are:

- a.) turmeric and litmus
- b.) phenolphthalein and methyl orange
- c.) litmus and methyl orange
- d.) phenolphthalein and litmus

Q5. The salt which will give an acidic solution on dissolving in water is:

- a.) KCl
- b.) NH_4Cl
- c.) Na_2CO_3
- d.) CH_3COONa

Q6. What are the observations when quick lime is added to water?

Q7. Which indicator gives pink colour in the basic solution?

20 Answers Acids And Bases

Monica M. Tighe, Marg Brown



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Fluid, Electrolyte and Acid-Base Disorders Alluru S. Reddi, 2023-05-09 This updated edition provides a clear and concise understanding of the fundamentals of fluid electrolyte and acid base disorders that are frequently encountered in clinical practice Each chapter follows a standard format that begins with pertinent basic physiology followed by its clinical disorder Cases for each fluid electrolyte and acid base disorder are discussed along with board type questions with explanations to increase clinicians knowledge Revised with new developments in the field this edition s expanded chapters cover useful information left out of other textbooks This practical current and clinically oriented book is a must have reference for practicing physicians students residents and fellows

TEXT BOOK OF PHARMACEUTICAL INORGANIC CHEMISTRY Dr. Ajay D. Kshirsagar; Ms. Pradnya A. Bhosle (Kshirsagar), 2025-04-01 The Textbook of Pharmaceutical Inorganic Chemistry is a comprehensive guide designed for students and professionals in the pharmaceutical sciences It covers essential concepts related to the role of inorganic compounds in medicinal chemistry and drug formulation The book begins with an exploration of impurities in pharmaceutical substances detailing their sources types and methods of detection using limit tests for elements like chloride sulfate iron arsenic lead and heavy metals The acids bases and buffers section provides insights into buffer solutions isotonicity and their pharmaceutical applications The book also delves into major electrolytes their physiological functions and their importance in replacement therapy including sodium chloride potassium chloride and calcium gluconate Dental products such as fluoride based treatments dentifrices and desensitizing agents are thoroughly discussed A significant portion focuses on gastrointestinal agents including acidifiers antacids cathartics and antimicrobials with detailed mechanisms and applications The book also examines expectorants emetics and haematinics with examples like potassium iodide ammonium chloride and ferrous sulfate Additionally the text

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ACID-Base, Fluid and Electrolytes Mr. Rohit Manglik, 2024-07-30 Detailed understanding of acid base homeostasis fluid distribution and electrolyte regulation in the human body essential for clinical diagnosis and treatment

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Fundamentals of Sustainable Chemical Science Stanley E. Manahan, 2009-03-10 Written by Stanley Manahan *Fundamentals of Sustainable Chemical Science* has been carefully designed to provide a basic introduction to chemistry including organic chemistry and biochemistry for readers with little or no prior background in the subject Manahan bestselling author of many environmental texts presents the material in a practical

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