Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

Service	zinc	iron (II)	iron (III)	gallium	silver	lead (IV)
chloride	ZnCl ₂	FeCl ₂	FeCl ₃	GaCl ₃	AgCI	PbCI ₄
acetate	Zn(C ₂ H ₁ O ₂) ₂	Fe(C ₂ H ₂ O ₂) ₂	Fe(C ₂ H ₃ O ₂) ₃	Ga(C ₂ H ₃ O ₂) ₃	Ag C ₂ H ₃ O ₂	Pb(C ₂ H ₁ O ₂) ₄
nitrate	Zn(NO ₃) ₂	Fe(NO ₃) ₂	Fe(NO ₃) ₃	Ga(NO ₃) ₃	AgNO ₃	Pb(NO ₃) ₄
oxide	ZnO	FeO	Fe ₂ O ₃	Ga ₂ O ₃	Ag ₂ O	PbO ₂
nitride	Zn ₃ N ₂	Fe ₃ N ₂	FeN	GaN	Ag ₂ N	Pb ₃ N ₄
sulfate	ZnSO ₄	FeSO ₄	Fe ₂ (SO ₄) ₃	Ga ₂ (SO ₄) ₃	Ag ₂ SO ₄	Pb(SO ₄) ₂

Write the formulas for the following compounds:

- copper (II) chloride CuCl₂
- lithium acetate LiC₂H₃O₂
- vanadium (III) selenide V₂Se₃
- 4) manganese (IV) nitride Mn₃N₄
- 5) beryllium oxide BeO
- sodium sulfate Na₂SO₄
- 7) aluminum arsenide AlAs
- potassium permanganate KMnO₄
- 9) chromium (VI) cyanide Cr(CN)₆
- 10) tin (II) sulfite SnSO₃
- 11) vanadium (V) fluoride VF₅
- ammonium nitrate NH₄NO₃

Answers For Ionic Formula Writing

Christopher R. Hren, John T. Moore, Peter J. Mikulecky

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