# EXPERIMENT 1: PERCENT SUGAR IN BUBBLE GUM

# **Data Sheet**

Table 2: Sugar (Sucrose) Data

Mass of unchewed gum (a)	Mass of chewed gum (g)	Mass of sugar (g)	Percent of sugar in gum
5.6	2.6	3	46.42
	Section to the Land		

# Post-Lab Questions

- 1. What is the percent sugar? Show your calculations. 2.6/5.6 x 100 = 46.42
- 2. What is the molar mass of sugar (sucrose, C12H22O11)? Show your calculations.
- 3. Convert the mass of dissolved sugar to moles. Show your calculations. I
- 4. Convert the moles of sugar to molecules of sugar. Show your calculations.
- Imagine that you used four pieces of gum instead of two. Calculate the number of moles.Show your calculations.
- 6. How does the mass of the saliva alter the experimental data?

# **Answer Percent Sugar In Bubble Gum Lab**

**Lei Huang** 

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