

GCSE Chemistry (AQA Higher Tier) Multiple Choice Questions

Unit: C1

Q#	Question	A	B	C	D
1	Which of these is the correct group number for Magnesium?	1	2	3	4
2	Which of these elements forms a 1+ ion when bonded ionically?	Mg	Na	Pb	Al
3	Which of these elements is in Group 3?	Al	Pb	Fe	Br
4	Which of these is the correct periodic table symbol for Gold?	Gu	Go	Au	Ag
5	Which ion is formed by Aluminium when it bonds ionically?	1+	2+	3+	4+
6	Which of these elements is liquid at room temperature?	Al	Fe	He	Br
7	Which group contains a series of unreactive gases?	4	7	8/0	3
8	Which of these elements, despite being in Group 1, does not have all the qualities of a metal?	He	H	Hg	Te
9	Which of these is the correct mass and charge of a neutron?	M:1 C:0	M:1 C+1	M:0 C:1	M:0 C+1
10	Three neutrons, two protons and one electron would have a cumulative charge equal to which of these?	+1	-1	+2	-2
11	Which of these is the correct approximate mass of an electron?	1/1200	1/1800	1/290	1/10

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B Lingard



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