Types of Chemical Reactions POGIL

Do atoms rearrange in predictable patters during chemical reactions?

why?

Recognizing patterns allows us to predict future behavior. Weather experts use patterns to predict dangerous storms so people can get their families to safety. Political analysts use patterns to predict election outcomes. Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions.

Model 1 - Types of Reactions

Set A	Set !
$4\text{Fe}(s) + 3O_{j}(g) \rightarrow 2\text{Fe}_{2}O_{j}(s)$	Mg0
$N_2(g) + 3H_2(g) \rightarrow 2NH_1(g)$	81.i,
$2SO_{s}(g) + O_{s}(g) \rightarrow 2SO_{s}(g)$	2H,
$MgO(s)+H_2O(l) \rightarrow Mg(OH)_2(aq)$	2800
$P_2O_3(g) + 3H_2O(l) \rightarrow 2H_2PO_4(aq)$	-2Na
$SO_3(g) + H_2O(l) \rightarrow H_2SO_4(aq)$	(NII
Set C	Set
$2FeCl_s(aq) + 3Zn(s) \rightarrow 2Fe(s) + 3ZnCl_s(aq)$	AgN
$2Al(NO_s)_s(aq) + 3Ca(s) \rightarrow 3Ca(NO_s)_s(aq) + 2Al(s)$	2143
$Mg(s) + CuSO_s(aq) \rightarrow MgSO_s(aq) + Cu(s)$	Nag
$2Al(s) + 6HCl(aq) \rightarrow 2AlCl_{s}(aq) + 3H_{s}(g)$	
$Cl_{r}(g) + 2NaBr(aq) \rightarrow 2NaCl(aq) + Br_{r}(l)$	FeS(
$ZnBr_2(aq) + F_2(g) \rightarrow ZnF_2(aq) + Br_2(l)$	HC

Set B	
$MgCO_{j}(t) \rightarrow MgO(t) + CO_{j}(g)$	
$8\text{Li}_2S(s) \rightarrow 16\text{Li}(s) + S_s(s)$	
$2H_{\downarrow}O(l) \rightarrow 2H_{\downarrow}(g) + O_{\downarrow}(g)$	
$2KClO_1(s) \rightarrow 2KCl(s) + 3O_1(g)$	
$2Na_sO_s(s) \rightarrow 2Na_sO(s) + O_s(g)$	
$(\mathrm{NH}_y)_2\mathrm{CO}_y(s) \rightarrow 2\mathrm{NH}_y(g) + \mathrm{H}_2\mathrm{O}(l) + \mathrm{CO}_2(g)$	
Set D	
$AgNO_s(aq) + NaCl(aq) \rightarrow AgCl(s) + NaNO_s(aq)$	ŀ.
2HNO,(aq) + Mg(OH),(aq) → Mg(NO),(aq) + 2H,O(l)	
$Na_{\mu}CO_{\mu}(aq) + CaCI_{\mu}(aq) \rightarrow CaCO_{\mu}(a) + 2NaCI_{\mu}(aq)$	
$FeS(s) + 2HCl(aq) \rightarrow H_sS(g) + FeCl_s(aq)$	
$HCl(aq) + NaOH(aq) \rightarrow H_2O(l) + NaCl(aq)$	
FeBr. (aq) + K.PO. (aq) \rightarrow FePO. (s) + 3KBr(aq)	

1. Match each description below to one of the reactions sets (A, B, C, or D) from Model 1.

______ Ionic compounds dissolved in water switch partners.
One compound breaks into elements or smaller compounds
Two or more elements or compounds combine to form one product.
Part of an ionic compound is removed and replaced by a new element.

- 2. Define the following terms as they are commonly used in the English language:
 - a. Synthesis:
 - b. Decomposition:
 - c. Replacement:
 - 3.The four sets of chemical reactions shown in Model 1 have the following general names. Discuss within your group which name belongs to which set of chemical reactions. Write the name in the appropriate place in Model 1.

Answer Key To Types Of Chemical Reactions

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Table of Contents Answer Key To Types Of Chemical Reactions

- 1. Understanding the eBook Answer Key To Types Of Chemical Reactions
 - The Rise of Digital Reading Answer Key To Types Of Chemical Reactions
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Answer Key To Types Of Chemical Reactions
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Types Of Chemical Reactions
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answer Key To Types Of Chemical Reactions
 - Personalized Recommendations
 - Answer Key To Types Of Chemical Reactions User Reviews and Ratings
 - Answer Key To Types Of Chemical Reactions and Bestseller Lists

- 5. Accessing Answer Key To Types Of Chemical Reactions Free and Paid eBooks
 - Answer Key To Types Of Chemical Reactions Public Domain eBooks
 - Answer Key To Types Of Chemical Reactions eBook Subscription Services
 - Answer Key To Types Of Chemical Reactions Budget-Friendly Options
- 6. Navigating Answer Key To Types Of Chemical Reactions eBook Formats
 - o ePub, PDF, MOBI, and More
 - Answer Key To Types Of Chemical Reactions Compatibility with Devices
 - Answer Key To Types Of Chemical Reactions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Types Of Chemical Reactions
 - Highlighting and Note-Taking Answer Key To Types Of Chemical Reactions
 - Interactive Elements Answer Key To Types Of Chemical Reactions
- 8. Staying Engaged with Answer Key To Types Of Chemical Reactions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Types Of Chemical Reactions
- 9. Balancing eBooks and Physical Books Answer Key To Types Of Chemical Reactions
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Answer Key To Types Of Chemical Reactions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To Types Of Chemical Reactions
 - Setting Reading Goals Answer Key To Types Of Chemical Reactions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To Types Of Chemical Reactions
 - Fact-Checking eBook Content of Answer Key To Types Of Chemical Reactions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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