

Grade 11 Chemistry

INTERMOLECULAR FORCES & Physical Properties

EXAM QUESTIONS PART 3



NAME OF SUBSTANCE	FORMULA	BOILING POINT (°C)
Water	H ₂ O	100
Ethanol	CH ₃ CH ₂ OH	78
Bromine	Br ₂	58,8
Iodine	I ₂	184,3
Ammonia	NH ₃	-33,3
Phosphine	PH ₃	-87,7

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

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LP Steffe



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