

15.2**HOMOGENEOUS AQUEOUS SYSTEMS****Section Review****Objectives**

- Distinguish between a solvent and a solute
- Describe what happens in the solution process
- Explain why all ionic compounds are electrolytes
- Demonstrate how to write the formula for a hydrate

Vocabulary

- | | | |
|--------------------|------------------|----------------------|
| • aqueous solution | • solvation | • strong electrolyte |
| • solvent | • electrolyte | • weak electrolyte |
| • solute | • nonelectrolyte | • hydrate |

Key Equation

- $\text{Percent H}_2\text{O} = \frac{\text{mass of water}}{\text{mass of hydrate}} \times 100\%$

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Water is a polar liquid and an excellent 1 for many substances. Aqueous solutions are 2 mixtures of ions or molecules in water. The solubility of a solute depends on solute-solvent interactions. A good rule to remember is 3.

Substances that dissolve as ions are known as 4. A solute that is completely ionized in solution is a 5 electrolyte. A weak electrolyte is only 6 ionized. A solution of an electrolyte will 7 an electric current, whereas a solution of a 8 is nonconducting.

Many crystals are 9; they contain water of hydration. In the process called 10, the water of hydration is lost from a hydrate that is exposed to air.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

Answers For Homogeneous Aqueous Systems Section Review

J. D. R. Thomas



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Water Science Reviews 4: Volume 4 Felix Franks, 1989-11-09 The fourth volume of Water Science Reviews presents three fascinating accounts of hydration phenomena in colloidal systems O F Evans and David Miller provide a reappraisal of the role of water in promoting amphiphilic assembly and structure Donald England's review of water soluble polymers highlights those areas that show unique solution properties or where there is contention as to the explanation for the behavior The final review by Kenneth Newman addresses the hydration of surfaces a topic of profound scientific and technological importance Post graduate researchers interested in topical critical reviews will benefit from this volume

Catalog of Nuclear Reactor Concepts Charles E. Teeter, James A. Lecky, John H. Martens, 1964

Nuclear Science Abstracts, 1965-05

Technology Quarterly and Proceedings of the Society of Arts, 1898 Vol 8 14 include Review of American chemical research edited by Arthur A Noyes

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