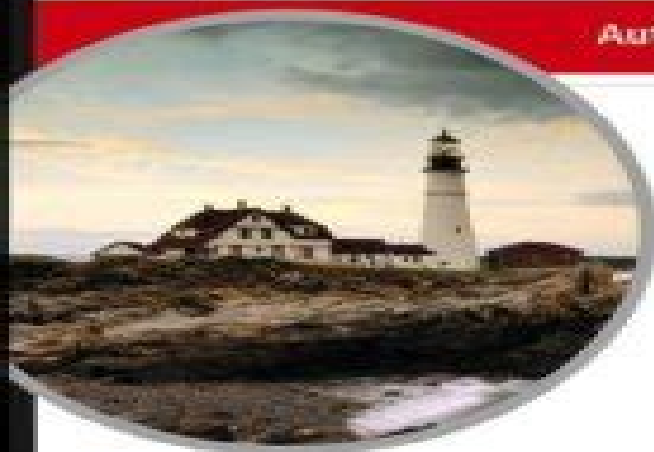


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in the n 2 molecule the nitrogen atoms have an σ bond and two π bonds holding the two atoms together the presence of three strong bonds makes n 2 a very stable molecule phosphorus is a third period element and as such does not form π bonds efficiently therefore it must fulfill its bonding requirement by forming three σ bonds

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