

Chemistry Worksheets Class 10 on Chapter 2 Acids, Bases and Salts - Set 2

Q1. Fresh milk has a pH of 6. When milk changes into curd, the pH value will:

- a.) become 7
- b.) become less than 6
- c.) become more than 7
- d.) remains unchanged

Q2. Which of the following is medicine for indigestion?

- a.) sodium hydroxide
- b.) manganese hydroxide
- c.) magnesium hydroxide
- d.) potassium hydroxide

Q3. How many number of water of crystallisation is present in copper sulphate crystals?

- a.) 3
- b.) 5
- c.) 7
- d.) None

Q4. The indicators which turn red in acid solution are:

- a.) turmeric and litmus
- b.) phenolphthalein and methyl orange
- c.) litmus and methyl orange
- d.) phenolphthalein and litmus

Q5. The salt which will give an acidic solution on dissolving in water is:

- a.) KCl
- b.) NH_4Cl
- c.) Na_2CO_3
- d.) CH_3COONa

Q6. What are the observations when quick lime is added to water?

Q7. Which indicator gives pink colour in the basic solution?

20 1 Answers Acids And Bases

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Physical Chemistry for Engineering and Applied Sciences Frank R. Foulkes, 2012-09-12 Physical Chemistry for Engineering and Applied Sciences is the product of over 30 years of teaching first year Physical Chemistry as part of the Faculty of Applied Science and Engineering at the University of Toronto Designed to be as rigorous as compatible with a first year student's ability to understand the text presents detailed step by step derivations of the equations that permit the student to follow the underlying logic and of equal importance to appreciate any simplifying assumptions made or mathematical tricks employed In addition to the 600 exercises and end of chapter problems the text is rich in worked non trivial examples many of which are designed to be inspiring and thought provoking Step by step derivation of all equations enables the student to smoothly follow the derivation by sight and can be understood relatively easily by students with moderate skills and backgrounds in mathematics Clear and accessible Physical Chemistry for Engineering and Applied Sciences includes The answers to all of the 112 worked examples 99 exercises following many of the worked examples and 496 end of chapter problems Topics not normally seen in introductory physical chemistry textbooks ionic reaction rates activities and activity coefficients or not regularly explained in much detail electrochemistry chemical kinetics with an eye on industrial applications Special appendices that provide detailed explanations of basic integration and natural logarithms for students lacking a background in integral calculus An in depth chapter on electrochemistry in which activities and activity coefficients are used extensively as required for accurate calculations *General Chemistry* Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value *Textbook of Medical Biochemistry - E-Book* Dinesh Puri, 2022-11-18 The fifth edition of this book is thoroughly revised and updated as per guidelines of NMC in accordance with the competency based curriculum of Biochemistry It focuses primarily on clarity of the fundamental concepts with a logical sequence of events that aids learning The organization of content in this profusely illustrated book provides the essential knowledge of biochemistry without extraneous details Authentic resource material for undergraduate medical students NEXT USMLE PLAB etc New to this Edition Addition of new chapter on Processing and Targeting of Proteins Systematically modified chapters on Cancer Recombinant DNA Technology and Genetic Engineering Molecular Biology Eukaryotic Gene Expression Immunology Organ Function Tests Carbohydrates and Lipid Metabolism and Energy Metabolism and Nutrition to cater to various competencies recommended by new curriculum Inclusion of clinical boxes in each chapter highlighting horizontal and vertical integration of topics to foster solid understanding Enriched text with additional new line diagrams clinical photographs tables and boxes for easy understanding and reproducibility Multiple choice questions have been given chapterwise to evaluate the level of

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basic scientific principles and appropriate medical diagnoses and treatment The authors have used a practical clinical approach to the problems throughout emphasising points of relevance and interest with illustrative case examples The book should appeal to clinicians emergency room staff biochemists laboratory technologists medical and science students

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